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**Handbook on measurement methods and strategies
at very low levels and activities**

P H Burgess

Abstract

Decommissioning of nuclear and other sites which process or employ radioactive materials is becoming increasingly important. This handbook suggests methods which can be used to assess the levels of radioactive contamination and neutron activation in materials which have the potential for free release, ie, those which are only, at worst, contaminated or activated to a low level.

**Radiation Metrology
National Radiological Protection Board
Chilton
Didcot
Oxon OX11 0RQ**

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1 Introduction

1.1 Objectives

Interest in decommissioning nuclear facilities is increasing, both in terms of work in progress and in terms of the long term planning for facilities which are expected to operate for many years or which are still in the course of construction and planning. Costs of disposal of active waste are rising and there is an increasing awareness of the value of recycling useful materials. These aspects encourage operators to classify with increasing care materials from equipment and site demolition and clearance.

Guidance on acceptable levels for the free release of materials is being developed within the European Community¹ (EC) for general application for both surface and bulk activity. This handbook

- (i) reviews the capabilities and limitations of various types of monitoring equipment and practicable applicable methods, which can be used to demonstrate compliance with standards set for release of material,
- (ii) provides generic cost considerations, and
- (iii) gives examples of the methods and instrumentation which have been employed and which look attractive for future work.

Projects considered have ranged in size from the recovery of limited amounts of relatively high value materials such as copper pipe to large scale projects such as the dismantling of a very large gaseous diffusion plant.

1.2 Economic background

Decommissioning can be a very difficult process, involving the handling of very heavily contaminated or activated items, some of which, such as asbestos, are themselves problem materials. These materials will not be considered in this handbook which will concentrate on the generally much larger volume of materials which are either very lightly contaminated or activated, or which may not be radioactive at all. Examples of the latter include structural steel and concrete from a reactor building which have never been exposed to a significant neutron flux, electrical cables, motors, air conditioning ducts from the clean side and many others. The operator may however either wish, or be obliged, to demonstrate that contamination or bulk activities allow free release of these materials.

The economics of such programmes are not always clear cut. Measurement inevitably implies costs, which can be considerable both for equipment and labour. The value of the recycled materials may be intrinsically low, and it is possible that potential buyers may well offer prices below the recognised market value because of an understandable reluctance to handle materials which have been in any way connected with radiation. For some materials, such as concrete and wood, reuse may be possible within the nuclear licenced site, if the site is large and developing. Transport costs involved in, for example, transporting scrap steel to a smelter which is prepared to handle lightly contaminated materials may be significant. Operators may well be concerned about an individual sorting through a large volume of materials which have been assessed on the basis of mean activity and finding a small object which is considerably more radioactive than the mean. This object may then be used, quite incorrectly, to cast doubt on the whole process. Companies buying scrap materials may well be

concerned by the possibility of either honest mistakes or malicious acts by persons involved in sentencing the waste. Against this, operators may well take the view that costs of disposal to landfill will probably rise as time goes on and may well find it difficult to get an agreed disposal cost per tonne from the disposal site operator to cover the duration of the project. For many of these materials balancing these costs and deciding on the best option is inevitably going to be difficult.

The adoption of measurement strategies and methods that are both cost efficient and provide a transparent quality characterisation of waste is therefore highly desirable.

2 Measurement types

There are two distinct types of measurement to be considered. One is the monitoring of surface contamination, ie, the measurement of radioactive materials which may be attached to the surface of another object. These radioactive materials may well be subject to relatively easy removal, leading to potential ingestion or inhalation by persons involved in handling, transporting and reprocessing the scrap. They may also directly irradiate workers by the emission of α , γ or energetic β radiation. As a result of recycling the radioactive material may be concentrated in the recycled material or, as is more likely, discharged to the atmosphere or concentrated in recycling waste. Assessment of surface contamination is thus an important part of the sentencing of decommissioned materials.

The other broad category is contamination in bulk, caused either by direct neutron activation of the material and its inherent impurities, or by the diffusion of radionuclides through the surface of the material to a significant depth. These contaminants are inherently less accessible, which has the advantage that they are less likely to be ingested or inhaled during recycling. They will, however, continue to irradiate workers and will also be concentrated in the recycled materials, discharged to atmosphere or concentrated in waste. They are also more difficult to measure because the geometry is less defined and because self shielding is inevitable.

The two types of measurement also differ in the sense that surface contamination monitoring can normally be performed on a relatively well defined area whereas the assessment of contamination in bulk usually implies a significant averaging volume.

3 Planning a measurement programme

There are two questions to be asked before transporting material from a decommissioning site for recycling:-

- (1) Is the material fit for recycling, ie, does the total activity per unit mass comply with recycling criteria? This can be composed of two elements, activity from surface contamination and activity in the mass of the material.
- (2) Is the material readily transportable to the recycling facility, if it is off the site, or do special transport arrangements have to be made? This will normally be dictated by the removable surface activity under the current (1990) IAEA Regulations for the Safe Transport of Radioactive Material² and in future, when implemented, the 1996 Regulations³. Material is defined as contaminated for β and γ activity in excess of 0.3 Bq cm^{-2} in the 1990 Regulations, changing to 0.4 Bq cm^{-2} in the 1996 Regulations, and for alpha activity in excess of 0.03 Bq cm^{-2} , changing to 0.04 Bq cm^{-2} . Special transport arrangements are required for activities in

excess of 4 Bq cm^{-2} (β and γ) and 0.4 Bq cm^{-2} (α). The consequences of these values is that some material which is acceptable for free release for recycling, particularly material contaminated by low energy β emitters such as ^{147}Pm , will have to be transported as radioactive material.

Producing answers to these questions is the basis of the measurement programme.

4 Readily available sources of information

The process of deciding on the various options for the material generated in decommissioning is complicated. The extra costs of monitoring to a level sufficient to justify free release have to be balanced against the costs of storage on site or off-site disposal. It is important to assemble as much information as is easily available as a first step. For those site operators whose plant is still operating and some way from decommissioning there is an opportunity to construct monitoring and recording programmes that will help in the decommissioning phase.

4.1 Existing radiation protection monitoring data

Monitoring information should be available throughout the life of the plant. Such monitoring will normally have mainly been biased towards the designation of working areas in terms of dose rate, air and surface activity levels. Measurements in areas where maintenance was performed regularly should be well documented. Inaccessible areas will not be documented at all. The information may well be quite basic, such as count rate from a particular type of beta counter, gross alpha count rate on wipes, gamma ambient dose equivalent rate or neutron dose equivalent rate.

4.2 Authorised discharge route monitoring

For some areas useful information may well be available, especially where there are authorised discharge routes. Stack discharges on power stations for the heating, ventilation and air conditioning systems will normally be monitored using high quality γ spectrometers, generating values of Bq m^{-3} for a wide range of nuclides. Liquid discharge routes will also be monitored in a similar manner.

4.3 Plant integrity monitoring

Monitoring may have also been performed to check on plant integrity. A good example is the secondary cooling circuit on PWRs where regular monitoring is undertaken to identify heat exchanger leaks or, on gas cooled reactors, the burst can detector system which is designed to identify fuel failure.

4.4 Activation calculations

Knowledge of the likely neutron exposure and of the composition of structural materials will enable the calculation, if sometimes only very approximately, of the levels of activation nuclides such as ^{60}Co and ^{54}Mn .

4.5 Process throughput

In areas where fuel fabrication, reprocessing or other chemical operations have been undertaken, the total throughput of radioactive material may be well known. Care has to be taken where the deposition varies dramatically between the various chemical species. A good example is

the dramatic tendency of ^{99}Tc from reprocessed uranium to plate out on the surface of fuel fabrication plant.

4.6 Possible problems

Impediments to this process of gathering relevant data are many. In old plant monitoring results may have been lost or many be intrinsically inadequate. Accidents may have taken place leading to a much larger release of activity into the system than was anticipated at the construction stage. Complicated parts of the plant such as pipe bends may have much higher contamination levels than would be expected from measurements in other areas where deposition is less likely. Plant may have been deliberately run in an unusual condition or may have been modified and components replaced at some unspecified time. Alloys of a different composition to those specified may have been used. The construction of the plant may differ from the plans available. The plant may have been operated under some form of security condition which may mean that full monitoring data is not available.

The older the plant, and the more unusual the plant, the poorer the information that is likely to be available in written records. It is important in these circumstances to trace as many of the workers as possible, including those who have left the plant or retired, in order to get as complete as possible picture of the real history of the plant.

5 Assessing current plant condition

No matter how good the information available on the plant before shutdown, it will still be important to conduct more monitoring before planning a decommissioning and disposal operation. This programme should be designed to give sufficient information to allow the design, at least in broad outline, of the disposal monitoring programme, including the level of investment in automatic monitoring equipment, the number of staff to be assigned to the monitoring programme, training needs and the development of disposal cases for discussion with the relevant authorities.

It should be designed to identify the presence of difficult to monitor nuclides, which may not have been important while the plant was running but which may contribute significantly to the total activity. Such nuclides include the low energy (~ 6 keV) electron capture nuclides generated by neutron activation in steel and the very low energy beta emitters such as ^{63}Ni and ^{241}Pu . Detailed monitoring should be concentrated in the area between the material which is blatantly too active, as indicated by dose rate measurements for example, and the material which should definitely be clean or only trivially contaminated.

The most effective approach is the removal of samples for analysis by gamma spectrometry and radiochemical analysis. This gives an early opportunity to establish 'fingerprints' for different materials and areas of the site, and the identification of monitoring strategies and instrumentation.

Action lists follow which describe these points in more detail.

5.1 Action list for equipment and procedures for waste assessment

5.1.1 Surface contamination

- (a) Identify likely contaminants from operating data.
- (b) Weight the contaminants using the free release levels.
- (c) Identify those which are likely to be important.

- (d) Look at the decay scheme.
- (e) Identify nuclides which are likely to be present and emit reasonably penetrating radiation eg, beta emitters with an E_{\max} in excess of 0.6 MeV.
- (f) Choose a suitable instrument ie, one which responds to likely dominating contaminants, has a suitable area and is sufficiently robust.
- (g) Identify contaminated areas.
Are there any problems with gamma background?
- (h) Sample contamination and analyse by γ spectrometry and radiochemical analysis for alpha and beta emitters with insufficient γ emissions.
- (i) Compare (h) with (a). Are the results in reasonable agreement with prediction? Is the radionuclide mix reasonably consistent, at least in terms of the nuclides which will influence the free release of the material? Are there problem nuclides, in the sense of nuclides which are important in terms of the release criteria but which are difficult to detect on the material under consideration? A good example would be steel which is rusty and has significant α contamination.
- (j) Decide on whether direct monitoring is possible for the material and the condition in which it exists. If direct monitoring is not possible would cleaning be possible? If it is, is the initial instrument appropriate? Would there be advantages in negotiating with any supervisory authority for an increase in any initial defined averaging area using the initial results for justification?
- (k) Decide on the most appropriate instrument, ie, one which provides the most appropriate balance of characteristics
 - : has good sensitivity over background.
 - : is not dependent on changes in levels of unimportant nuclides.
 - : is as robust as possible having taken account of the demands above.
 - : is simple to operate by the staff selected.
 - : can be tested easily.
 - : is easy to repair.
 - : has an appropriate averaging area balanced against cost, complexity of the shape of the material to be monitored and the permitted averaging area.
- (l) Decide on the number of instruments required. Negotiate a repair service or identify a member of the team who can maintain the equipment. Order sufficient spares, having estimated the likely damage rate and considered the delivery time for supply of components from the manufacturer or component supplier.
- (m) Organise training for the workforce on both the instrument chosen and the monitoring technique.
- (n) Write monitoring and maintenance procedures.
- (o) Set up an auditing process, whereby results are traceable to national standards and also where a random sample of the result is checked by another competent and independent person.
- (p) Decide on the frequency of sampling for γ spectrometry and radiochemical analysis in order to support the calculated release level.
- (q) Start monitoring for release.
- (r) Take an early piece of the material which is contaminated at or around the release level. Ask all the workforce to monitor it, concealing, as far as possible, that others will have monitored

it. Compare the results. Is the spread acceptable? Would significant errors have occurred in the sense that either:

- (i) a piece which was definitely over the acceptable level was marked for free release or
 - (ii) a piece which was definitely within the limit was marked as excessively contaminated. If so, find out why the error occurred and take steps to prevent it, such as further training or a modification to a written procedure.
- (s) Continue the process bearing in mind the need to continue to be confident that the instrument indication corresponding to free release is unchanging. The process should be reviewed to ensure that samples are being sorted correctly.

5.1.2 Bulk contaminated and neutron activated materials

- (a) Identify likely contaminants from operating data.
- (b) Identify a hand held instrument which will respond to those contaminants.
- (c) If at all possible remove samples for analysis, using the hand held instrument to aid selection.
- (d) Analyse samples by γ spectrometry and by radiochemical analysis for alpha and beta emitters with insufficient γ emissions.
- (e) Weight the measured activities using the free release levels.
- (f) Identify those nuclides which are likely to be important.
Look at the decay scheme for each nuclide.
- (g) Are the samples reasonably consistent?
- (h) Choose, if possible, a useful emission, such as an energetic γ line, which is present at a level proportional to the weighted activity levels in the samples.
In the absence of one useful energy, is there a limited combination of gamma emissions which could be used to assess the waste?
- (i) In the absence of useful γ emissions is there some other possible means of monitoring such as the measurement of gross alpha or gross beta emissions from prepared samples?
- (j) Can the monitoring be performed in situ? Can the monitoring be performed after the materials have been removed but are intact, such as steel beams? Will monitoring have to be performed on what is essentially debris, such as concrete rubble?
- (k) Identify a suitable monitoring technique or techniques. It may well be advantageous to split the materials into 3 groups, the obviously clean, based on in situ measurement, which can go directly for free release, the obviously excessively active, again based on in situ measurement, and the borderline group. Materials in this group may require more sophisticated monitoring using installed equipment.
- (l) Identify suitable equipment to operate the monitoring techniques chosen.
- (m) Decide on the number of instruments required. Organise maintenance. Order sufficient spares, having estimated the likely damage rate and considered the delivery time for the supply of components from the manufacturer or component supplier.
- (n) Organise training for the workforce on the instruments and techniques chosen.
- (o) Write monitoring and maintenance procedures.
- (p) For installed monitoring, identify building needs and power requirements. As examples a sodium iodide scintillator conveyor monitoring system for crushed rubble will require a large but fairly basic building whereas a drum monitoring system using large intrinsic germanium

detectors will require a building which offers a good environment and the provision of a liquid nitrogen supply.

- (q) Set up an auditing process, whereby results are traceable to national standards and also where a random sample of the results are checked by another competent and independent person.
- (r) Decide whether there is a requirement for more detailed analysis of samples, by, for example, germanium detector spectrometry and/or radiochemical analysis. If so, at what frequency?
- (s) Start monitoring for release.
- (t) In the case of in situ monitoring ask all the work force to monitor at defined positions concealing, as far as possible, that others will have monitored at the same position. Compare the results. Is the spread acceptable? Would significant errors have occurred in the sense that either:
 - (i) a piece which was definitely over the acceptable level was marked for free release or
 - (ii) a piece which was definitely within the limit was marked as excessively contaminated. If so, can written procedures be improved or is extra training required?

Continue the process bearing in mind that the operator has to be confident that significant changes in radionuclide composition are identified and that the release criteria are adjusted accordingly.

6 Measurement of surface contamination

6.1 Introduction

Measurement of surface contamination is an important aspect of the decommissioning of nuclear facilities. Much of the effort is devoted to protecting the workforce and avoiding unacceptable discharges to the environment. However, for many materials which have not encountered significant neutron exposure and which thus cannot have been activated, contamination monitoring can be used for clearance of the materials. The method is particularly suitable for materials like stainless steel which, for the vast majority of nuclides, can be deemed to be a non-absorbing surface. Other metals fall into the same category, but materials like concrete and wood pose problems because of the effects of absorption into the surface. Other problems include the presence of paint on the surface which may have been applied for protection but which also may have been applied to fix loose contamination. This was a common technique in the past.

Contamination monitoring is particularly suitable for large flat surfaces but can also be adopted for surfaces like the inside of pipes and ducts if the shape of the detector is suitable.

6.2 Choice of detector

The choice of detector is determined by the radionuclides present, the ease of access to the surfaces to be monitored, the permissible maximum monitoring area, the presence or absence of a significant γ background and the skills of the workforce. The detector chosen obviously has to respond effectively to the radionuclides of interest, but in many cases an additional constraint is to avoid

responding to other nuclides which may be present but which are of much lower radiological significance. An example of this is ^{239}Pu , a radionuclide where acceptable levels are very low, which may be found in the company of beta emitting fission products, where the acceptable levels may be much higher.

The majority of relevant radionuclides can be detected by direct surface contamination monitoring but some radionuclides of significance, such as ^3H , ^{241}Pu and ^{63}Ni , all of which are low energy beta emitters, are very difficult to detect directly. Tritium also has a very strong tendency to become absorbed into surfaces, even of metals such as stainless steel, which makes surface monitoring unreliable.

The subsequent sections discuss the instrumentation, physics of operation, practical limitations and minimum detectable activities for different classes of radionuclides.

6.2.1 Alpha emitters

The main characteristics of alpha emissions are their limited disintegration energy range, typically 4 MeV to 8 MeV, the very high rate of energy loss, and the resultant extremely short range, typically of the order of 5 mg cm^{-2} or 5 cm in air.

This means that practical detectors have to be held within 10 mm of the surface under investigation. However, the very high rate of energy loss means that alpha particle detectors can be produced with a background of less than 1 count per minute for monitoring areas of 100 cm^2 . The other problem with the very short range is that even a very thin surface coating, such as 1 coat of gloss paint or a smear of oil or grease can reduce the emissions from the surface effectively to zero. Hence, alpha monitoring can only be used either where an apparently clean surface is available, such as the inside of a glove box, or where the activity is, to a degree, uniformly mixed, such as in soil. It is not useful for painted, dirty or porous surfaces, which will require cleaning before monitoring.

6.2.1.1 Scintillation counters for α monitoring

The scintillation counter is a popular detector for α monitoring. The scintillator is almost always a thin layer of zinc sulphide activated with silver. Zinc sulphide is an efficient scintillator in terms of light output per unit energy deposited, but it absorbs its own light. However, this is not a problem given that the crystals only need to be a few tens of microns thick to absorb completely the energy of the alpha particle. The zinc sulphide crystals are normally dispersed on a plastic sheet and are protected by a light tight window normally of aluminised melinex of 1 to 2 mg cm^{-2} total thickness. Some detectors use a direct coating of aluminium on the surface of the scintillator assembly, which is difficult to manufacture but which is extremely robust in service.

The size of the scintillator normally ranges from circular ones with an area of 20 cm^2 up to rectangular ones with an area of 600 cm^2 . The scintillator is mounted in a probe which comprises a case, the inside of which is painted white, into which protrudes a photomultiplier tube. This apparently crude arrangement, which relies on the multiple reflection of the light generated by the scintillation event, works well because of the very high light output per event. The maximum size is limited by a loss of efficiency of light collection from the corners of the scintillator. However, the maximum area found in service, 600 cm^2 , is probably the maximum size that makes operational sense, given typical averaging areas.

The photomultiplier is connected to a high voltage supply normally between 600 volts and 1500 volts. The pulse that emerges from the anode of the photomultiplier tube is fed to a discriminator,

which reacts to the large α pulses and ignores the smaller ones generated by x, γ photons and beta particles, and then either to a ratemeter or a counter timer.

Typical detection efficiencies for thin layer surface contamination are up to 40%, ie, 40% of the α particles which strike the window are counted, giving typical responses of $20 \text{ s}^{-1} \text{ Bq}^{-1} \text{ cm}^2$ for the 100 cm^2 counter and $100 \text{ s}^{-1} \text{ Bq}^{-1} \text{ cm}^2$ for the 600 cm^2 counter while background count rates in well designed, constructed and maintained equipment can be as low as 1 count per minute per 100 cm^2 ^(4,5). Minimum detectable activities thus tend to be limited by statistical uncertainties in the α count over a particular period, rather than by the influence of background count rate.

The majority of detectors are planar. However, there would be no particular problem in manufacturing counters which could have windows formed to a particular radius in order to allow the efficient monitoring of pipes or process vessels.

One particular constraint is the necessity to have the scintillator no more than a few mm from the surface in question. This can lead to problems in balancing the need to have the detector close enough to a surface to monitor it effectively while avoiding contamination of the detector. For decommissioning applications where large areas of flat surfaces or large areas of pipe work are to be monitored then it might well be worth equipping the detector with an automatic scanning mechanism which holds the detector at a fixed distance from, and which tracks it in a reproducible manner over, the surface. For smaller areas then detectors equipped with wheels may offer advantages especially in situations where there is a strong likelihood that contamination is negligible. Regular checking of the wheels for accumulated activity is essential. For complicated areas the skill and interest of the operator become paramount, and it becomes difficult to guarantee the quality of the programme.

Problems associated with scintillation counters are usually dominated by window damage, especially when the objects to be monitored have complicated or spiky shapes. It is essential to choose any protection to be fitted over the window with care. The grille should have openings which are sufficiently small to prevent the objects being monitored reaching the aluminised plastic window, sufficiently rigid to avoid being deformed onto the window and sufficiently strong to avoid tearing. The grille should also be sufficiently transparent to allow the monitoring target level to be achieved reliably. The most satisfactory design uses a thin metal foil which is etched into an open pattern. Woven wire grilles can appear quite open but have very poor transparency when viewed at oblique angles and are generally much less satisfactory. The grille should also be spaced two or three mm from the window to avoid the grille being forced back onto the window by chance contact with a raised part of the object being monitored.

Scintillation counters are also unsuitable for use in significant magnetic fields such as those associated with motors, some tools and sometimes even structural steel beams. The magnetic field deflects the electrons in their paths within the photomultiplier tube, reducing the gain and bringing the α generated pulses below the counter threshold. As a rather subjective guide, a medium sized screwdriver which is sufficiently magnetic to allow a chain of 4 steel paper clips to hang from its end is sufficiently magnetised to cause problems.

Prolonged rough use can also lead to problems with the zinc sulphide crystals detaching themselves from the scintillator plate. These crystals can then abrade the aluminising on the inside of the plastic foil leading to spurious background counts or unreliable operation in strong light.

Scintillation counters may also not be fail safe when the window is punctured. The counter can cease to respond to α activity without showing an increase in background count rate. Well designed instruments should detect a significant light leak as the current from the photomultiplier tube will rise above the normal operational level. This current can be used to trigger a fault warning which will alert

the system or operator. The window can then be replaced which, on a well designed detector, is a simple operation taking only a few minutes.

Operation in high dose rates produced by low energy γ emitters is also inadvisable. Dose rates of a few hundred μSv per hour can depress the gain of the photomultiplier without generating a high background count rate^{4,5}. This can be a problem when dealing with glove boxes which have been used for ²⁴¹Am. The reduction in gain causes the α pulses to fall below the counting threshold.

6.2.1.2 *Proportional counters for α monitoring*

Proportional counters are popular for α surface contamination monitoring. The detector is extremely simple in construction. Typically they are in the form of a shallow box with one of the large sides formed from a sheet of aluminised plastic, similar to that used for scintillation counter windows but with a lower necessary degree of light tightness. The other five sides are generally aluminium. Across the mid-plane of the detector parallel to the window is an array of thin parallel wires, normally tungsten or stainless steel about 25 μm in diameter. These anode wires are maintained at a potential of about +2 kV with respect to the counter walls and window. The counter is either used as a flow counter, in which counting gas, normally P10 (90% argon, 10% methane) flows through the counter at the rate of a few cm^3 per minute, as a refillable counter, which is refilled at intervals of a few hours with butane, or as an air counter, where dry air is used as a counting gas. This last is unusual in that air is a very poor counting gas. However, the very high rate of energy loss of an α particle of greater than 1 MeV cm^{-1} generates a sufficiently high charge density to allow detection.

The form of construction allows very large counters, which can have high length to width ratios. It is generally possible to obtain detectors which match the permissible averaging area and of a length to width ratio which can be optimised for the objects to be monitored. The output from the detector is similar to that from the scintillation detector in that it is proportional to the energy deposited but not proportional to the energy of the incident particle. Typical detectors have depths of approximately 2 cm, which gives energies deposited for α particles crossing the detector at right angles to the window of about 3 MeV and for energetic β particles of about 100 keV. The counting threshold can be set to distinguish against β radiation even for β particles travelling virtually in the plane of the anode wires. In a similar way to the scintillation detector some of the α particles will produce pulses below the maximum β pulses. These are particles which strike the window at shallow angles. Typical window thickness is 1 to 2 mg cm^{-2} , and hence particles incident at less than 20° to the window are unlikely to penetrate the window with sufficient energy to be counted.

Again, similar to scintillation counters, detection efficiencies of 50% for particles striking the window are possible giving sensitivities of 25 $\text{s}^{-1} \text{Bq}^{-1} \text{cm}^2$ for a 100 cm^2 counter^{4,5}. Detection efficiency tends to be much more uniform with a much less pronounced deterioration at edges and corners. Background count rates are also low, again of the order of 1 per minute per 100 cm^2 in a well maintained instrument. Minimum detectable activities are again limited by statistical uncertainties in the α count over a particular period.

In a similar way to scintillation detectors there are no serious problems in producing detectors with windows formed to a defined radius and which have a 360° view, barring the window supports and structural stiffeners. Such detectors can be made in diameters down to 15 mm, which allow monitoring of the inside of Zircalloy reactor fuel cladding tubes, for example.

The constraints on source to detector window spacing and grille transparency are common to all α detectors. However, flow proportional counters are slightly more resistant to damage. Tiny holes which can render scintillation counters unusable can be tolerated. Significant leaks can lead to a loss

of response for the volume of the detector down stream in the gas flow direction. The slight positive pressure in the counter also makes their windows much less susceptible to damage when compared to Geiger Muller tubes, which have a much higher negative pressure difference to sustain. Proportional counters are also much less susceptible to magnetic fields because the velocity of the electrons is very much lower. They can thus be used for monitoring articles with significant magnetic fields.

Window repair for the planar counters is very simple, usually involving removal of the window frame, replacement of the aluminised plastic and replacement of the window frame. The plastic can then be tightened by careful heating by a hot air blower. The anode wires are remarkably resilient given that they are invisible in most lighting conditions. Damage to the anode wires can result in total failure of the counter or a failure of a segment of the counter. Repair requires skill.

Other than obvious window damage the main problem tends to be chemical contamination of the counting gas. Low flow rates or tiny leaks can lead to a rise in oxygen concentration which results in a reduction in gain. The wrong feed piping can also cause problems. Flexible plastic piping can leach significant levels of plasticiser into the gas, resulting in loss of gain, or requiring excessive gas flow rates for satisfactory function.

For refillable types, best performance is obtained with continuous use and regular refilling. If a counter is not refilled for a long period then the purging process can take many minutes until oxygen is driven from the detector.

Air filled types also require desiccation, and, even with desiccation, may be unreliable in damp climates.

6.2.1.3 Geiger Muller counters

Geiger Muller counters are not normally suitable for α monitoring at free release levels because of their limited window areas and their high background count rates, not because of a low counting efficiency. The mass per unit area of a good quality mica windowed detector is similar to that of scintillation and proportional counters and hence the counting efficiency is similar, up to 40%, giving sensitivities of up to $4 \text{ s}^{-1} \text{ Bq}^{-1} \text{ cm}^2$ for a 20 cm^2 counter^{4,5}. The limited window area is a consequence of the low internal pressure and the subsequent stress on the window and the high background is a consequence of the inability to distinguish between events depositing different energies.

6.2.1.4 Semiconductor detectors

Semiconductor detectors are commonly found in α in air monitors where their ability to perform α spectrometry is useful in distinguishing plutonium from radon progeny, for example. These detectors have a thin entrance window, and areas up to 60 cm^2 . They have some use in surface monitoring as it is possible to make a detector of very limited thickness. This can be compared with the relative bulk of the photomultiplier used in the scintillation detector and the minimal depth of several mm required to give effective α separation from β , γ events in the proportional counter. Detection efficiency can be up to 30%, giving sensitivities of the order of $9 \text{ s}^{-1} \text{ Bq}^{-1} \text{ cm}^2$.

This detector differs from the scintillation, proportional and Geiger Muller counters in that there is no gain in the detector. The signal generated is purely that produced directly by the incident radiation. It thus requires a relatively low noise preamplifier. After amplification it is easy to set a threshold in a similar way to the scintillation and proportional counters to give a very clear discrimination against beta and gamma radiation. The lack of gain means that careful precautions have to be taken against electromagnetic interference, with very effective screening around detector and preamplifier. This results in detectors generally being fitted with aluminised plastic windows.

Modern examples are relatively robust and can be cleaned with care. Curved detectors can be produced using an array of relatively small detectors which can be connected in parallel. Operating voltages are less than 100 volts, which is a significant advantage over the other detectors, but their main advantage, exploited to the full in the monitoring of environmental samples, for example, is their very low α background.

6.2.1.5 Ionisation chamber detectors

A limitation of the previous 4 types of detectors described is the need to be very close to the surface under investigation, which effectively restricts them to flat or smoothly curved surfaces. The monitoring of complicated or inaccessible surfaces such as the inside of valves is not feasible. However, it is possible to monitor such objects by detecting the ionisation generated in air blown through or over the object. This allows the detection of α activity at ranges up to at least 6 metres⁶. Discrimination against β and γ radiation is generally good, although not up to the level of the more conventional techniques. Again, this is a consequence of the very high rate of energy deposition of an α particle which can deposit several MeV when emitted inside a 30 mm diameter pipe, whereas a β particle will normally deposit approximately 30 keV.

Detection efficiency is such that it is possible to detect levels of $0.02 \text{ Bq cm}^{-2} \alpha$ in a 1 metre length of 30 mm diameter pipe, in a γ background of $0.2 \mu\text{Sv h}^{-1}$.

The technique involves blowing air over or through the object to be monitored. The air is then passed through an ionisation chamber and the charge collected. The current generated per α particle depends on the local air velocity but above a particular level the signal becomes only slightly dependent on flow rate and, more importantly, only slowly dependent on distance of the source from the detector. The limitations of the process is that debris must not be blown into the ion chamber. Debris will inevitably carry an electric charge and will gradually build up on insulators producing high leakage currents and thus increasing the minimum detectable activity. The technique is thus particularly suitable for confirming that the insides of apparently clean process pipe work or unirradiated fuel cladding is in fact uncontaminated.

6.2.1.6 Summary of minimum detectable activities of α contamination for the different detector types, sizes and integrating periods

The minimum detectable activity is defined here as the activity which will be identified on 50% of measurements as being in excess of the background count rate at the 95% confidence level, ie, for a background sample, only 1 measurement in 40 will generate a false positive signal.

This is tabulated in Table 1 for the detector types described previously and for different effective integration times. Manual α contamination monitoring is generally based on the audio pulse output from an instrument rather than on the indicated count rate. The one second and 3 second columns are appropriate to this and correspond to rates of movement over a single active particle of the order of 10 cm s^{-1} and 3 cm s^{-1} for a typical 100 cm^2 detector and 30 cm s^{-1} and 10 cm s^{-1} for a 600 cm^2 detector. For the scintillation, proportional and semiconductor detectors an average of 1 count per second has been taken as significant for the one second survey time and 1 count in 3 seconds for the 3 second survey time. This is based on the observation that a skilled user will normally pause, and wait a few seconds, at any point where a count has occurred. The longer times are appropriate either to automatic methods or to fixed position integrating measurements where the area to be monitored is covered by placing the detector at one spot, integrating and then moving the detector to the adjacent area.

TABLE 1 Typical minimum detectable activities for α radiation (Bq cm⁻²)

Effective counting time(s)	Minimum detectable activity (Bq cm ⁻²)						
	Scintillation detectors		Proportional counters		Geiger Muller detectors	Semiconductor detectors	Ion* chamber detectors
	100 cm ²	600 cm ²	100 cm ²	1000 cm ²	20 cm ²	60 cm ²	2200 cm ²
1	0.05	0.01	0.05	0.005	0.6	0.1	0.01
3	0.015	0.002	0.015	0.002	0.3	0.03	0.005
10	0.012	0.002	0.012	0.001	0.16	0.02	0.003
30	0.007	0.001	0.007	<0.001	0.09	0.015	0.002
100	0.004	<0.001	0.004	<0.001	0.05	0.01	0.001

*In the shape of a pipe 2.4 m long and 34 mm in internal diameter.

6.2.1.7 Summary of minimum measurable activities of α contamination for the different detector types, sizes and integrating periods

In this case it is not the detection of the presence of activity but the ability to measure its level with a reasonable degree of precision that is estimated. For this case this will be assumed to correspond to a 95% confidence level of $\pm 50\%$ of the mean value. For pulse counting integrating detectors with insignificant background count rates this corresponds to a total of 16 counts in the monitoring period. For the Geiger Muller detector the high background is also taken into account. For the 3 second value the use of a ratemeter with a 3 second time constant was assumed while for the longer times an integrating measurement was assumed. The results are shown in Table 2.

TABLE 2 Typical minimum measurable α activities (Bq cm⁻²)

Effective counting time(s)	Minimum measurable activity (Bq cm ⁻²)					
	Scintillation detectors		Proportional counters		Geiger Muller detectors	Semiconductor detectors
	100 cm ²	600 cm ²	100 cm ²	1000 cm ²	20 cm ²	60 cm ²
3	0.1	0.03	0.1	0.01	0.9	0.3
10	0.08	0.016	0.08	0.008	0.7	0.2
30	0.03	0.006	0.03	0.003	0.3	0.06
100	0.01	0.002	0.01	0.001	0.2	0.02

The minimum measurable activity for the ion chamber system is of the order of 20 Bq, total.

6.2.2 β emitters

β emitters are characterised by the wide ranges of maximum energies found in practice from 10 keV to over 3 MeV and by the fact that each nuclide emits a range of energies from close to zero up to a maximum, with the average energy being at about 30% of maximum energy. The lower energy nuclides, such as ³H and ²⁴¹Pu, are impossible to monitor directly using counters with windows and can only be monitored by swabbing or by special windowless counters. On the other hand nuclides such

as ^{90}Y and ^{106}Rh have ranges in air of several metres and penetrate 1 cm thick plastic sheets to a significant degree. Hence, for some nuclides, beta monitoring is restricted to effectively clean surfaces with the probe mounted within a few mm of the surface, whereas for other nuclides direct monitoring is possible even for painted surfaces.

The main problem often with direct β monitoring is the presence of a γ background, sometimes from the material itself and sometimes from other active materials in the environment. Instruments have been designed which reject, to a degree, γ background, by comparing, for example, the count rate from two superimposed detectors both of which see virtually the same γ field but only one of which sees the β radiations. These are uncommon in surface contamination monitoring at the moment and will not be considered further.

6.2.2.1 Scintillation counters for β monitoring

Scintillation counters for β monitoring generally use the same basic form as those for α monitoring. Most comprise a flat thin piece of plastic which either contains the scintillant or which is coated on one side by scintillating crystals. This is protected by a thin light tight window. On some designs with relatively small areas the scintillator is coupled to the photomultiplier tube by a light guide but in the majority of cases the scintillator plate is mounted in a probe which comprises a light tight case the inside of which is painted white and into which protrudes the photomultiplier tube. A proportion of the light from the scintillator will reach the photomultiplier directly but much will be reflected at least once by the inside of the case. Again it is surprising how well this arrangement works in practice.

There is a large variety of scintillators which are encountered in practice. Detectors designed to monitor low energy β emitters such as ^{35}S and ^{14}C generally use anthracene, which is very efficient. This is in the form of small crystals. Scintillators normally have thickness of only a few mg cm^{-2} as this minimises the response to x, γ radiation. It also has the interesting effect of leading to energy depositions in the scintillator for high energy β particles which are less than those generated by β particles of energies of 100 keV. The window is again normally formed of aluminised plastic with thicknesses in the range 1 mg cm^{-2} to 3 mg cm^{-2} . Some detectors intended for high energy emitters use aluminium sheet windows with thicknesses similar to cooking foil.

Areas range generally from 20 cm^2 to 600 cm^2 , limited by a loss of efficiency of light collection from the corners of the scintillator.

For β detectors the counting threshold is set close to the point at which thermoelectric noise from the photomultiplier sets in, which for anthracene scintillators corresponds to energy depositions of approximately 20 keV. The signal is then fed to a ratemeter, normally with an audio output, for manual use or to a scaler timer for automatic operation. The scintillator is normally flat but there are some which are formed around a central light guide and, in principal, there is no reason why cylindrical or curved detectors could not be built for pipe monitoring etc.

Detection efficiencies and corresponding responses are given in Tables 3 and 4 derived from reference 4 for a range of types and sizes. Detection efficiency is defined as the probability of a count for a particle striking the window of the detector.

Response is defined as the count rate (s^{-1}) for a uniform source greater in dimension than the detector with an activity of 1 Bq cm^{-2} at a distance of 3 mm.

The probes considered are all manufactured by Bicron NE Limited (or their predecessors) but are comparable with those produced by other manufacturers.

TABLE 3 Typical detection efficiencies (β scintillation probes)

Type	Area (cm ²)	Detection efficiency (%)			
		High energy β (⁹⁰ Sr+ ⁹⁰ Y)	Medium energy β (³⁶ Cl)	Low energy β (¹⁴ C)	α (²³⁸ Pu)
BP4	20	55	51	29	36
BP6/4A	100	46	44	17	32
*DP2	49	27	17	–	22
*DP3	100	34	26	–	34
*IDP6AD	100	44	42	–	37
BP17	600	38	33	No data	28

*Dual phosphor probes combining a zinc sulphide layer on a plastic scintillator sheet. The responses are obtained using the beta and alpha channels on a dual ratemeter.

TABLE 4 Typical responses (β scintillation probes)

Type	Area (cm ²)	Response (s ⁻¹ Bq ⁻¹ cm ²)			
		High energy β (⁹⁰ Sr+ ⁹⁰ Y)	Medium energy β (³⁶ Cl)	Low energy β (¹⁴ C)	α (²³⁸ Pu)
BP4	20	5.5	5.1	2.9	3.6
BP6/4A	100	23	22	8.7	16
*DP2	49	6.8	4.3	–	5.4
*DP3/4A	100	17	13	–	17
*IDP6AD	100	22	21	–	18.5
BP17	600	114	100	No data	84

*Dual phosphor probes combining a zinc sulphide layer on a plastic scintillator sheet. The responses are obtained using the beta and alpha channels on a dual ratemeter. Note that when correctly set up, only genuine α pulses should appear in the α channel. However, a proportion of the α events will appear in the β channel.

The response to γ background is also relevant. This is quoted in Table 5 in terms of counts s⁻¹ μ Gy⁻¹ h air kerma for ¹³⁷Cs γ radiation incident normal to the probe window under conditions of secondary electron equilibrium. The background count rate in an area of low normal background is also given.

TABLE 5 Typical background count rates (scintillation detectors)

Probe	Normal level (s ⁻¹)	Response to ¹³⁷ Cs γ radiation (s ⁻¹ μ Gy ⁻¹ h)
BP4	2	5
BP6/4A	4	25
DP2	2	3
DP3/4A	3	45
DP6	4	25
BP17	28	250

The levels of γ response are very variable between probe types and depend critically on the setting of the counting threshold. For most designs the response drops rapidly as the photon energy decreases. This is illustrated in Figure 1. The corresponding curves for proportional and Geiger Muller counters tend to rise, often sharply, as the energy decreases. This can be an important point in favour of the use of scintillation detectors when working in areas contaminated either by low energy nuclides or where radiations from higher energy sources such as ¹³⁷Cs or ⁶⁰Co have been multiply scattered.

Problems with scintillation counters are generally dominated by window damage, as discussed for α probes, but in this case the normal manifestation is an increase in background count rate in artificially lit areas. Magnetic fields can again be a problem and long term rough use can also cause the scintillant crystals to part company from their support plate or light guide leading to a loss of efficiency

and abrasion of the metal coating on the window. The detectors are also extremely variable with a range of operating voltages within any type of a factor of 2.

6.2.2.2 Proportional counters for β monitoring

Proportional counters are popular for β monitoring. The general design is the same as that for α monitoring, with a relatively thin window and an array of anode wires. There are, however, two forms which differ in window material and gas fill. One form is the sealed type, which generally use xenon as a counting gas, and has a titanium window about 5 mg cm^{-2} thick. These have a low α sensitivity but a relatively high sensitivity to low energy photons, caused by the high Z, high density gas fill. The other form is basically identical to the α detector, and is generally found in dual α/β monitoring instruments ie, it has a thin plastic window and is either a flow counter or is refilled with counting gas, normally P10 (90% argon, 10% methane) or butane. Air does not work for beta detection. This second variety can be produced with windows formed to a radius and which have a 360° view, barring structural stiffeners. Areas range from hand held units, usually with 100 cm^2 to 200 cm^2 windows, up to units of 1000 cm^2 for mounting in installed equipment or for use on vehicles for road monitoring. Detection efficiencies and corresponding responses are given in Tables 6 and 7 for a range of types and sizes^{4,5}.

TABLE 6 Typical detection efficiencies (β proportional counters)

Type	Area (cm ²)	Fill gas	Detection efficiency (%)			
			High energy β (⁹⁰ Sr+ ⁹⁰ Y)	Medium energy β (³⁶ Cl)	Low energy β (¹⁴ C)	α (²³⁸ Pu)
BZ100XEP	100	Xenon	46	44	9	18 ⁽¹⁾
BZ200XK-P	200	Xenon	45	40	11	18 ⁽¹⁾
MZ100	100	Butane	56	56	46	44 ⁽²⁾
H1370W	165	Butane	58	55	41	36 ⁽²⁾

- (1) No α , β separation.
- (2) Instrument with α , β separation, α data taken from α channel. Note that in a correctly adjusted unit only α events appear in the α channel. However a proportion of the α events will appear in the β channel.

TABLE 7 Typical responses (β proportional counters)

Type	Area (cm ²)	Fill gas	Response (s ⁻¹ Bq ⁻¹ cm ²)			
			High energy β (⁹⁰ Sr+ ⁹⁰ Y)	Medium energy β (³⁶ Cl)	Low energy β (¹⁴ C)	α (²³⁸ Pu)
BZ100XEP	100	Xenon	23	22	4.5	9 ⁽¹⁾
BZ200XK-P	200	Xenon	45	40	11	18 ⁽¹⁾
MZ100	100	Butane	28	28	23	22 ⁽²⁾
H1370W	165	Butane	48	45	34	30 ⁽²⁾

- (1) No α , β separation.
- (2) Instrument with α , β separation, α data taken from α channel. Note that in a correctly adjusted unit only α events appear in the α channel. However a proportion of the α events will appear in the β channel.

The response to γ background is also relevant. This is quoted in Table 8 in counts $\text{s}^{-1} \mu\text{Gy}^{-1} \text{h}$ (air kerma) for ^{137}Cs γ radiation incident normal to the probe window under conditions of secondary electron equilibrium^{4,5}. The background count rate in an area of low normal background is also given.

TABLE 8 Typical background count rate (proportional counters)

Detector	Normal level (s^{-1})	Response to ^{137}Cs γ radiation ($\text{s}^{-1} \mu\text{Gy}^{-1} \text{h}$)
BZ100XEP	6	50
BZ200XK-P	10	98
MZ100	5	35
H1370W	5	40

The γ responses are very consistent within each type. For the xenon filled types the response rises rapidly as the energy decreases, as illustrated in Figure 2. For the butane filled types the increase in response is much less marked, as illustrated in Figure 3. Proportional counters can be at a disadvantage compared to scintillation counters of the same sensitivity in areas where there is significant low energy x, γ radiation.

Problems have already been discussed in the section on α detectors, and mainly revolve around window damage. This can be difficult to detect and is often best achieved by immersing the detector in water, taking care to protect electrical connections. A leak shows a stream of tiny bubbles. The detector should have been filled to its normal working pressure in order to avoid water entering the counter.

6.2.2.3 Geiger Muller detectors for β contamination monitoring

Geiger Muller detectors have many uses in β contamination monitoring where either small areas are to be monitored or where there are areas which are inaccessible to larger detectors. Detectors for β contamination monitoring fall into two classes, mica windowed types which can operate at low energies (^{14}C , ^{35}S) and thin metal or glass walled types, which are useful only for medium and high energy β emitters.

Mica windowed detectors have window thicknesses normally in the range 1 to 3 mg cm^{-2} and areas from about 1 cm^2 up to a maximum of 20 cm^2 . The maximum window size is limited by the stress on the mica caused by the low pressure (10% of atmospheric pressure) inside the detector. European designs operate generally at approximately 500 volts while most designs from the USA operate at 900 volts. Most are halogen quenched, and hence have effectively unlimited lives, but some organic quenched types can be found which have lives limited to about 10^8 to 10^9 counts.

Steel and glass walled detectors generally have wall thicknesses of about 30 mg cm^{-2} , and operate over a range of voltages. Again both halogen quenched and organic quenched designs are used. Because the walls are thicker and the materials stronger than mica, it is possible to make much larger sizes. Before the advent of scintillation and proportional counters, G-M detectors up to 1 metre long and 60 mm in diameter were commonly available.

Unlike scintillation and proportional counters the pulse size is independent of the radiation causing it. No energy selection or α/β discrimination is possible electronically. However the relatively simple electronics required to power the detector and detect the subsequent pulses are especially useful to designers of hand held equipment.

Typical detection efficiencies and responses are given in Tables 9 and 10 for a range of types and sizes based on data from reference 4.

TABLE 9 Typical detection efficiencies (β Geiger Muller detectors)

Type	Window or wall material	Thickness (mg cm ⁻²)	Area (cm ²)	Detection efficiency (%)			
				High energy β (⁹⁰ Sr+ ⁹⁰ Y)	Medium energy β (³⁶ Cl)	Low energy β (¹⁴ C)	α ²³⁸ Pu
ZP1481	Mica	1.5	3	27	20	7	14
ZP1430	Mica	2	6	63	60	30	30
DN212	Mica	1.5	20	48	43	26	34
7313	Mica	2	20	65	65	24	30
B6	Glass	35	15	19	8	0	0

TABLE 10 Typical responses (β Geiger Muller detectors)

Type	Window or wall material	Thickness (mg cm ⁻²)	Area (cm ²)	Responses (s ⁻¹ Bq ⁻¹ cm ²)			
				High energy β (⁹⁰ Sr+ ⁹⁰ Y)	Medium energy β (³⁶ Cl)	Low energy β (¹⁴ C)	α ²³⁸ Pu
ZP1481	Mica	1.5	3	0.4	0.3	0.1	0.2
ZP1430	Mica	2	6	1.9	1.8	0.9	0.9
DN212	Mica	1.5	20	4.8	4.3	2.6	3.4
7313	Mica	2	20	6.5	6.5	2.4	3.0
B6	Glass	35	15	1.4	0.6	0	0

The response to γ background is also important in many circumstances. This is quoted in Table 11 in counts s⁻¹ μ Gy⁻¹ h (air kerma) for ¹³⁷Cs γ radiation incident normal to the probe window under conditions of secondary electron equilibrium. The background count rate in an area of low normal background is also given.

TABLE 11 Typical background count rate (Geiger Muller detectors)

Detector	Normal level (s ⁻¹)	Response to ¹³⁷ Cs γ radiation (s ⁻¹ μ Gy ⁻¹ h)
ZP1481	0.2	1.8
ZP1430	0.3	2.5
DN212	1.0	6.5
7313	0.7	5
B6	0.6	5

The energy response tends to rise as the x, γ energy decreases, reaching a maximum at 60 keV (Figure 4). The ratio of response at 60 keV to that at 662 keV (¹³⁷Cs) ranges from 4 to 10 depending on type.

Geiger Muller detectors are normally very consistent within each type. Individual adjustment of polarising potential is not required and the manufacturers specified value can be used with confidence. This is an advantage during repair. The main problems with Geiger Muller detectors are their extreme vulnerability to window damage and the impossibility of repair of the detector. The mica windows are very thin and highly stressed. Even gentle contact with anything in the least pointed will cause their collapse. The only option, then, is replacement. Contrast this with the relative ease with which scintillation detector and refillable proportional counter windows can be replaced. Sealed proportional counters can normally be repaired by the manufacturer. However the cost of a Geiger Muller detector instrument is lower than that of scintillation and proportional counter instruments.

6.2.2.4 *Solid state detectors for beta contamination monitoring*

Large area silicon diode detectors have been used for direct contamination monitoring but the application is unusual. The main reason is the problem of detecting low energy events in a detector with a large area and corresponding high capacitance. The detector has no internal gain and this makes an instrument using such detectors vulnerable to radio frequency interference, especially as only very thin screening windows are acceptable where low energy beta particles are to be detected. Such detectors tend, then, to be found in applications such as β in air monitoring where they can be installed in a well screened housing.

6.2.2.5 *Ion chamber detectors for beta contamination monitoring*

There is no parallel to the technique used for α monitoring. The main reasons are the generally much lower energy per event for a β disintegration compared to an α and the much longer range of the β particle which may well escape the monitoring volume. For these reasons the response to β emissions of an α monitoring ion chamber system is typically less than 1% of the α response, for objects such as pipes.

6.2.2.6 *Summary of minimum detectable activities for β contamination for different detector types, sizes and integrating periods*

The minimum detectable activity is defined here as the activity which will be identified on 50% of measurements as being in excess of the background count rate at the 95% confidence level ie, for a background sample only 1 measurement in 40 will generate a false positive signal.

This is tabulated in Tables 12 and 13 for the detector types described previously, for different integration times and for energetic ($^{90}\text{Sr}+^{90}\text{Y}$) and low energy (^{14}C) β emitters. Manual β contamination monitoring is generally based on the audio pulse output from an instrument rather than on the indicated count rate. The 1 second and 3 second columns are appropriate to this. Background count rates are much higher than for α monitoring, even for the smallest detectors. The limits are derived from either a count rate in excess of background at the 95% confidence limit or 3 counts per second, whichever is higher for the one second period, and a count rate in excess of background at the 95% confidence limit or 2 counts per second, whichever is higher, for the 3 second period.

The longer times are appropriate either to automatic methods or to fixed position integrating measurements where the area to be monitored is covered not by continuous movement of the probe but by placing the detector at one spot, integrating, and then moving the detector to the adjacent area.

TABLE 12 Typical minimum detectable activities (background dose rate) for energetic β emitters ($^{90}\text{Sr}+^{90}\text{Y}$)

Effective counting time(s)	Minimum detectable activity (Bq cm ⁻²)					
	Anthracene scintillation detectors		Proportional counters		Geiger Muller detector	
	100 cm ²	600 cm ²	100 cm ²	100 cm ²	20 cm ² mica	15 cm ² glass
			xenon	butane		
1	0.2	0.1	0.2	0.2	0.5	2
3	0.1	0.05	0.1	0.1	0.3	1.4
10	0.06	0.03	0.07	0.05	0.1	0.3
30	0.03	0.02	0.04	0.03	0.05	0.2
100	0.02	0.01	0.02	0.01	0.03	0.1

TABLE 13 Typical minimum detectable activities (background dose rate) for low energy β emitters (^{14}C)

Effective counting time(s)	Minimum detectable activity (Bq cm ⁻²)				
	Scintillation detectors		Proportional counters		Geiger Muller detector
	100 cm ²	600 cm ²	100 cm ²	100 cm ²	20 cm ² mica
			xenon	butane	
1	0.5	No data	0.6	0.2	1.3
3	0.3	No data	0.3	0.1	0.8
10	0.15	No data	0.16	0.06	0.2
30	0.1	No data	0.1	0.03	0.1
100	0.05	No data	0.06	0.02	0.07

Frequently, however, measurements may have to be performed in significant γ dose rates, caused either by activation of the object being monitored or by other local sources. This will inevitably increase the background count rate and make the detection of surface contamination more difficult. In such cases the surface can be monitored with the detector window covered by an absorber to remove the beta particles and then the exercise repeated without the absorber and the indications compared. The increased count rate will not normally allow the easy detection of activity using the ear. It will be necessary to use the instrument indication. The minimum detectable activity is then that which corresponds to two standard deviations of the instrument indication. The calculation below is based on a dose rate of $5 \mu\text{Gy h}^{-1}$, ^{137}Cs γ radiation, giving a count rate of $N \text{ s}^{-1}$, and an instrument time

constant of 3 seconds. The corresponding standard deviation $(\sigma) = \left(\frac{N}{2T} \right)^{\%o}$. Hence the minimum

detectable activity is that that gives a count rate equal to $2 \left(\frac{N}{2T} \right)^{\%o} = \left(\frac{2N}{T} \right)^{\%o}$

. This is the calculation

used for the 3 second counting interval. For the higher times the calculation was based on an integrated

count over the stated time, ie, it is the activity which produces a count rate equal to $2\sqrt{\frac{T}{N}}$.

The results are given in Table 14.

TABLE 14 Typical minimum detectable activities at 5 $\mu\text{Gy h}^{-1}$ for energetic β emitters ($^{90}\text{Sr} + ^{90}\text{Y}$)

Effective counting time(s)	Minimum detectable activity (Bq cm^{-2})						
	Anthracene scintillation detectors		Proportional counters		Geiger Muller detector		
	100 cm^2	600 cm^2	100 cm^2	100 cm^2	20 cm^2 mica	15 cm^2 glass	
			xenon	butane			
3	0.4	0.3	0.6	0.4	0.6		3
10	0.3	0.2	0.4	0.3	0.5		2
30	0.2	0.1	0.3	0.2	0.3		1
100	0.1	0.06	0.1	0.1	0.2		0.7

6.2.2.7 *Summary of minimum measurable activities for β contamination for different detector types, sizes and integrating periods*

In this case it is not the detection of the presence of contamination but the ability to measure its level with a reasonable degree of precision that is estimated. This is assumed to correspond to a 95% confidence level of $\pm 50\%$ of the mean value. Again in this case the measurement is based on a ratemeter with a time constant of 3 seconds and for an integrating measurement for the longer times. Allowance is made for the influence of background count rate. For the ratemeter based measurement

$$\sigma = 0.25s = \left(\frac{s+b}{2T} \right)^{50}$$

where

σ = standard deviation

s = count rate from contamination (s^{-1})

b = background count rate (s^{-1})

T = response time (s)

$$\frac{s+b}{2T} = \frac{s}{16} \quad 16(s+b) = 2Ts^2$$

For $T=3$

$$\therefore 6s^2 = 16(s+b)$$

$$3s^2 - 8s - 8b = 0$$

For the integrating measurement

$$\sigma = 0.25st = ((s+b)t)^{1/2}$$

where t is the integrating time

$$s^2 t^2 = 16t(s+b)$$

$$t = \frac{16(s+b)}{s^2}$$

This analysis assumes that the background b is well known. This gives the results in Tables 15 and 16 for measurements at background dose rates.

TABLE 15 Typical minimum measurable activities (background dose rate) for energetic β emitters ($^{90}\text{Sr} + ^{90}\text{Y}$)

Effective counting time(s)	Minimum measurable activity (Bq cm ⁻²)					
	Anthracene scintillation detectors		Proportional counters		Geiger Muller detector	
	100 cm ²	600 cm ²	100 cm ²	100 cm ²	20 cm ² mica	15 cm ² glass
			xenon	butane		
3	0.2	0.1	0.2	0.2	0.5	2.3
10	0.15	0.01	0.2	0.1	0.3	1.5
30	0.08	0.04	0.09	0.07	0.1	0.6
100	0.04	0.02	0.05	0.04	0.04	0.3

TABLE 16 Typical minimum measurable activities (background dose rate) for low energy β emitters (^{14}C)

Effective counting time(s)	Minimum measurable activity (Bq cm ⁻²)					
	Anthracene scintillation detectors		Proportional counters		Geiger Muller detector	
	100 cm ²	600 cm ²	100 cm ²	100 cm ²	20 cm ² mica	
			xenon	butane		
3	0.5	No data	1.0	0.3	1.4	
10	0.4	No data	0.8	0.2	0.8	
30	0.2	No data	0.5	0.1	0.3	
100	1	No data	0.3	0.05	0.1	

6.2.3 x, γ emitters

Many radionuclides, or their immediate decay products, emit x, γ radiation as well as α or β radiation. In most circumstances it is best to monitor for these nuclides using the α or β emissions as the radiations striking the detector have to come mainly from the adjacent surface, whereas γ radiations can penetrate through the walls of the detector, unless it is extremely well shielded, and also through the object being monitored.

However, there is a limited number of nuclides which do not emit useable α or β radiations. Examples include ^{51}Cr , ^{55}Fe , ^{57}Co and ^{125}I . Some α emitting nuclides also emit significant x-rays,

typically L x-rays in the 13 to 20 keV region. Monitoring of these x-rays can provide useful information on the levels of α activity under paint, for example. Generally the minimum detectable activity will be much in excess of that which can be achieved using the α or β radiations if they were useable but it may be adequate for the end point required.

Two types of instrument are generally used for monitoring low energy x, γ radiations. They are the thin sodium iodide scintillation detector, fitted with a beryllium window, and the xenon filled proportional counter, which is generally fitted with a 5 mg cm⁻² titanium window. Both types of equipment are inherently capable of measuring x, γ radiation down to 5 keV and both are available with suitable areas for surface contamination monitoring.

Three examples will be discussed here, as illustrations. Two are sodium iodide scintillation detectors, and one is a xenon filled proportional counter. Both are suitable for connection to gated counting equipment or multichannel analysers, which helps to reduce the background count rate and hence the minimum measurable and detectable activities. Electron capture nuclides often have complicated decay schemes, emitting a range of energies with different probabilities. Hence the most effective manner for describing the operation of these detectors is to calculate the detection efficiencies over a range of photon energies. These can be folded with the energy and emission characteristics of the nuclide in question and the area of the detector to give a response.

Table 17 describes the detectors in more detail and gives their calculated detection efficiencies based on virtual contact with a thin source covering the window area.

The background count rate is also given for a simple counter, ie, one with a low energy threshold set at approximately 4 keV and no additional energy selection.

TABLE 17 Typical low energy x, γ detection efficiencies

Detection medium	sodium iodide	Sodium iodide	Xenon gas
Area (cm ²)	10	127	100
Thickness (mm)	3	2	20
Window material	beryllium	aluminium	titanium
Thickness (mg cm ⁻²)	47	14	5
Background count rate (s ⁻¹)	6.5	40	6
Overall detection efficiencies (counts photon ⁻¹ incident on the window)			
E (keV)			
5	0.7	0.02	0.01
10	0.9	0.6	0.25
20	1	0.9	0.16
30	1	1	0.06
40	1	1	0.12
50	1	1	0.08
60	1	1	0.05
100	0.9	0.8	0.02
200	0.56	0.22	0.01
300	0.12	0.18	<0.01

Gating the detector to the relevant photopeak will generally lead to a reduction in background by a factor of 5 up to 60 keV for the sodium iodide detectors.

6.2.4 Calculations of free release levels for surface contamination by a mix of nuclides

The preceding sections have described the performance of a range of detector types for individual nuclides. In most situations decommissioning will involve dealing with mixtures of nuclides, with a range of energies. As has been discussed the most useful concept is the finger print, a comprehensive analysis of the nuclides present. A limiting surface activity can be calculated for free release of the materials in the following way.

Fraction of release limit represented by 1 Bq cm⁻² of the mixture.

$$= \sum_A \frac{NA}{TA}$$

where NA = fraction of activity of nuclide A in the finger print, (Bq (A)/Bq (Total))

TA = limiting value for nuclide A, (Bq cm⁻²).

$$\text{Maximum permissible surface activity} = \left(\sum \frac{NA}{TA} \right)^{-1}$$

As an example consider the following mixture

Nuclide	Fraction of activity	Limiting value (Bq cm ⁻²)
A	0.3	1
B	0.3	10
C	0.4	100

$$\sum \frac{NA}{TA} = \frac{0.3}{1} + \frac{0.3}{10} + \frac{0.4}{100} = 0.334$$

$$\text{Maximum permissible activity (Bq cm}^{-2}\text{)} = (0.334)^{-1} \\ \approx 3$$

This calculation has to be followed up by a calculation of the response of the chosen monitor to the mixture. Typical test and calibration data for a beta monitor will give only 5 useful points, typically ¹⁴C, ¹⁴⁷Pm, ⁶⁰Co, ³⁶Cl or ²⁰⁴Tl, ⁹⁰Sr + ⁹⁰Y. Calculation of the response for the nuclides found in practice will require a degree of interpolation.

The first step is to obtain the decay scheme for each nuclide from a reliable reference such as ICRP Publication 38⁽⁷⁾. This should be weighted using the activity fractions derived from the finger print to produce an expected decay spectrum per unit finger print activity.

The second step is to calculate the instrument's likely response to these nuclides. The simplest approach is probably to graph the measured instrument responses against maximum β energy and then read off the response for the energies of interest. There is a degree of uncertainty associated with this

but a degree of uncertainty is inevitable, given that the β decay spectral shapes vary and given that self absorption from real contamination will inevitably lead to an under response for low energy nuclides.

Consider the following example. The finger print indicates the presence of 3 nuclides, in the stated proportions with the corresponding emissions.

Nuclide	Proportion	Emission	
		Energy (MeV)	Proportion
A	0.3	1	0.75
		0.3	0.25
B	0.3	0.20	1.00
C	0.4	0.015	1.00

Hence, per Bq 'finger print' the emissions are

Energy (MeV)	Proportion		
1	0.3×0.75	=	0.225
0.3	0.3×0.25	=	0.075
0.2	0.3×1.00	=	0.3
0.015	0.4×1.00	=	0.4

For practical direct monitoring nuclide C, which is tritium (^3H), is not detectable. The remaining emissions are detectable and the values can be read off the graph, giving a total response of $0.225 \times 10 + 0.075 \times 4 + 0.3 \times 2 = 3.15 \text{ s}^{-1} \text{ Bq cm}^{-2}$.

The maximum acceptable activity is 3 Bq cm^{-2} , giving a limit of count rate above background of $3 \times 3.15 \text{ s}^{-1} = 9.45 \text{ s}^{-1}$. Given the uncertainties inevitable in this process then a rejection level of 9 to 10 s^{-1} above background would be appropriate for thin layer contaminants.

This process illustrates the point that in some circumstances it may be better to choose an instrument with a restricted response because the final answer will be less susceptible to self absorption. If, for example, an instrument had been chosen with a negligible response to nuclides with a maximum β energy below 0.3 MeV then the response would fall but the end result would be less susceptible to contamination thickness fluctuation.

For x, γ emitters, such as ^{55}Fe , the same process should be followed, except that the decay scheme is generally more complicated. However it is likely that these emitters will not be limiting in most situations and reliance can be placed on beta monitoring to control the release level.

6.3 Testing and calibration of monitoring equipment

Periodic testing and calibration of monitoring equipment can normally follow the schemes used for routine radiation protection equipment⁽⁸⁾ and should be performed at suitable intervals, eg, annually. In essence testing should confirm that the instrument is fit for the use intended.

For surface contamination monitors this usually implies:-

- (1) confirmation that the response to large area sources of the type and energy range which the detector is designed to measure and expected to encounter is close to that specified by the manufacturer.
- (2) that the instrument has a suitably low response to radiation types and energies which it is designed not to detect. For example α surface contamination monitors should have a response to energetic β contamination of less than 1% of that to α contamination and also should have a negligible change in characteristics when exposed to 662 and 60 keV γ radiations at levels up to 1 mGy h⁻¹.
- (3) that the instrument is reasonably linear over its expected operating range i.e. that the ratio of the instrument indication to the value of the quantity of interest for a fixed radiation type and energy does not vary seriously.
- (4) that the response of the instrument is reasonably uniform over the full area of the instrument and that there are no large areas of greatly reduced performance.
- (5) that the instrument is in a good state of repair, ie, there are no obvious faults which are likely to cause a falling off in performance.
- (6) that the instrument does not respond to and is not inhibited by, for example, strong light levels or magnetic fields. This last category is very dependent on the exact circumstances of use.

Testing should be performed in a suitably equipped facility staffed by suitably qualified and knowledgeable personnel. All measured responses should be directly traceable to national standards and the estimated levels of uncertainty should be such that the field measurements are not significantly compromised. Testing should be to a written protocol and clearly documented.

If repairs have been performed on instruments which could seriously alter their performance, they should be checked in the same manner. Examples are replacement of a photomultiplier tube or a xenon filled proportional counter. Minor repairs, such as the replacement of a knob or the repair of a battery terminal would require only a simple function check.

Function checks should be performed regularly, especially given that many decommissioning measurements will be performed in difficult conditions. For each instrument a test should be devised which is demanding but yet rapid and which can be performed by the actual instrument user or the person who issues the instruments. As an example, for a typical anthracene scintillation detector a check of background count rate and a check with a ¹⁴C contamination plaque will detect the majority of faults. Again these tests should be to a simple written procedure. Careful watch has to be kept on equipment used for decommissioning to identify equipment which is becoming unacceptably contaminated. The classic case is an α monitor where low level contamination can cause the background to rise from 1 count per minute to 1 count every three or four seconds. This level can compromise the detection of areas of unacceptable contamination.

6.4 The thickness of typical surface coatings and their influence on contamination monitor performances

There is a very limited amount of data on the thickness of paint, varnish, polish and grease layers found in practice. A short series of measurements was undertaken to estimate the effect, using pieces of 50 mm square 1 mm thick aluminium plate. These were degreased, weighed and then coated in the way described. After 4 days the plates were reweighed and the mean mass per unit area of the

coating determined. This mass per unit area was then reproduced using thin sheets of aluminised plastic and the reduction in net count rate for a range of detectors and nuclides determined. The coatings are described in Table 18 and then their influence estimated in Table 19.

Reference 9 gives additional data for the attenuation characteristics for a range of nuclides, detectors and materials.

TABLE 18 Measured coating thicknesses

Material	Number of applications	Mass per unit area (mg cm ⁻²)
Car spray paint	One coat	2.6
Paint (anti rust)	One, according to manufacturer's instructions	4.0
Lacquer (for cars)	One coat	2.5
Wood varnish	One coat type to manufacturer's instructions	1.4
Furniture polish (beeswax)	Two	0.1
Oil (Castrol GTD)	One with saturated tissue	1.3
	Wiped down with a clean tissue	0.14
Grease (Castrol LM)	One with a saturated tissue	1.8
	Wiped down with a clean tissue	0.12

TABLE 19 Influence of coatings

Nuclide	Detector	Transmission %								
		Paint (car)	Paint (anti-rust)	Lacquer	Wood varnish	Furniture polish	Oil as applied	Wiped off	Grease as applied	Wiped off
²³⁸ Pu (α)	DP3	10	0	10	30	90	30	90	20	90
¹⁴ C (Low E β)	EP15	50	30	50	60	95	70	95	60	95
³⁶ Cl (medium β)	EP15	92	88	88	95	100	95	100	95	100
⁹⁰ Sr+ ⁹⁰ Y (energetic β)	EP15	95	93	95	100	100	100	100	100	100
⁵⁵ Fe (5.9 keV x ray)	LB1210B	30	35	50	70	97	70	95	60	95

The DP3 is a 100 cm² dual phosphor α , β detector, used on the α channel.

The EP15 is a 20 cm² thin mica window Geiger Muller detector

The LB1210B is a 100 cm² xenon filled proportional counter

The conclusion that can be drawn is that monitoring of the lower range nuclides, in this case ^{238}Pu , ^{14}C and ^{55}Fe , is extremely dependent on surface condition.

6.5 Summary

6.5.1 Transport

Table 20 below gives an indication of the ability of the instruments specified to detect contamination at acceptable levels for unrestricted transport⁽²⁾. In the future, levels will change to 0.4 Bq cm^{-2} (β, γ) and 0.04 Bq cm^{-2} (α)⁽³⁾ which will not change the markings.

TABLE 20 Compliance with transport limit

Radionuclide	Main emissions	Main energy (keV) (E _{max} for β s)	Detectable at a level of 0.3 Bq cm^{-2} (β, γ) or 0.03 Bq cm^{-2} (α) using portable equipment					
			100 cm ² anthracene scintillator	100 cm ² xenon proportional counter	100 cm ² butane proportional counter	20 cm ² Geiger Muller detector	100 cm ² zinc sulphide scintillator	100 cm ² thin sodium iodide scintillator
^3H	β	17	–	–	–	–	–	–
^{14}C	β	156	√	–	√	–	–	–
^{54}Mn	γ	835	–	–	–	–	–	√
^{55}Fe	x	6	–	–	–	–	–	√
^{60}Co	β	318	√	√	√	–	–	–
^{59}Ni	x	6	–	–	–	–	–	√
^{63}Ni	β	66	–	–	–	–	–	–
^{65}Zn	x, γ	9,1111	–	√	√	–	–	√
^{90}Sr	β	546	√	√	√	√	–	–
(^{90}Y)	β	2284	√	√	√	√	–	√
^{94}Nb	β	471	√	√	√	√	–	–
^{99}Tc	β	292	√	√	√	√	–	–
^{106}Ru	β	3600	√	√	√	√	–	–
$^{108\text{m}}\text{Ag}$	x, γ	24, 430-720	–	–	–	–	–	√
$^{110\text{m}}\text{Ag}$	β	–	–	–	–	–	–	√
^{125}Sb	β	612	√	√	√	√	–	√
^{134}Cs	β	660	√	√	√	–	–	–
^{137}Cs	β	511	√	√	√	–	–	–
^{147}Pm	β	224	√	–	√	–	–	–
^{151}Sm	β	76	–	–	–	–	–	–
^{152}Eu	β, x, γ	Numerous	–	–	–	–	–	√
^{154}Eu	β, x, γ	Numerous	√	√	√	–	–	–
^{238}U	α	Numerous	–	–	√	–	√	–
^{237}Np	α	Numeous	–	–	√	–	√	–
^{238}Pu	α	5500	–	–	√	–	√	–
^{239}Pu	α	5156	–	–	√	–	√	–
^{240}Pu	α	5168	–	–	√	–	√	–

Radionuclide	Main emissions	Main energy (keV) (E _{max} for βs)	Detectable at a level of 0.3 Bq cm ⁻² (B,γ) or 0.03 Bq cm ⁻² (α) using portable equipment					
			100 cm ² anthracene scintillator	100 cm ² xenon proportional counter	100 cm ² butane proportional counter	20 cm ² Geiger Muller detector	100 cm ² zinc sulphide scintillator	100 cm ² thin sodium iodide scintillator
²⁴¹ Pu	β	21	–	–	–	–	–	–
²⁴¹ Am	α	5486	–	–	√	–	√	–
²⁴⁴ Cm	α	5805	–	–	√	–	√	–

6.5.2 Clearance levels

The analysis below in Table 21 is based on the clearance levels proposed in reference 1 and describes the ability of a typical instrument in each category to demonstrate compliance with the clearance level for each nuclide for the direct reuse of metal items.

TABLE 21 Compliance with recycling limits

Radionuclide	Clearance level Bq cm ⁻²	Detectable at the clearance level					
		100 cm ² anthracene scintillator	100 cm ² xenon proportional counter	100 cm ² butane proportional counter	20 cm ² Geiger Muller detector	100 cm ² zinc sulphide scintillator	100 cm ² thin sodium iodide scintillator
³ H	1 x 10 ⁴	–	–	–	–	–	–
¹⁴ C	1 x 10 ³	√	√	√	√	–	–
⁵⁴ Mn	10	–	–	–	–	–	–
⁵⁵ Fe	1 x 10 ³	–	√	√	√	–	√
⁶⁰ Co	1	√	√	√	–	–	–
⁵⁹ Ni	1 x 10 ⁴	–	√	–	√	–	√
⁶³ Ni	1 x 10 ³	√	–	√	√	–	–
⁹⁰ Sr	10	√	√	√	√	–	√
⁹⁴ Nb	1	√	√	√	√	–	–
⁹⁹ Tc	1 x 10 ³	√	√	√	√	–	–
¹⁰⁶ Ru	10	√	√	√	√	–	–
^{108m} Ag	1	–	–	–	–	–	√
^{110m} Ag	1	–	–	–	–	–	–
¹²⁵ Sb	10	√	√	√	√	–	√
¹³⁴ Cs	1	√	√	√	√	–	–
¹³⁷ Cs	10	√	√	√	√	–	–
¹⁴⁷ Pm	1 x 10 ³	√	√	√	√	–	–
¹⁵¹ Sm	1 x 10 ³	√	–	√	√	–	–
¹⁵² Eu	1	–	–	√	–	–	√
¹⁵⁴ Eu	1	√	√	√	√	√	–
²³⁸ U	1	–	–	√	√	√	–

0.1

0.1

0.1

0.1

10

0.1

Radionuclide	Clearance level Bq cm ⁻²	Detectable at the clearance level					
		100 cm ² anthracene scintillator	100 cm ² xenon proportional counter	100 cm ² butane proportional counter	20 cm ² Geiger Muller detector	100 cm ² zinc sulphide scintillator	100 cm ² thin sodium iodide scintillator
²³⁷ Np		–	–	√	–	√	–
²³⁸ Pu		–	–	√	–	√	–
²³⁹ Pu		–	–	√	–	√	–
²⁴⁰ Pu		–	–	√	–	√	–
²⁴¹ Pu		–	–	–	–	–	–
²⁴¹ Am		–	–	√	–	√	–
²⁴⁴ Cm		–	–	√	–	√	–

The majority of nuclides are capable of detection at the levels appropriate for unrestricted transport and for unrestricted recycling.

Problem nuclides are confined mainly to the very low energy β emitters such as ³H and ²⁴¹Pu.

No one instrument is suitable for monitoring every nuclide. The butane filled proportional counter is probably the most versatile but is the only instrument considered which requires regular maintenance, in the shape of refilling with counting gas.

In most situations, then, choice of monitor will be determined by initial finger printing of the contamination. The actual monitoring limit will generally have to be calculated in counts per second taking into account nuclide mix and monitor characteristics, and, in some cases, may well be dominated by a relatively low fraction of a particular radionuclide which has a very high detection efficiency. A typical example would be ⁹⁰Sr and its daughter, ⁹⁰Y, both of which are very easy to detect but which are acceptable, for recycling, at a relatively high level.

7 Measurement of Specific Activity

7.1 Introduction

There is increasing interest in the recycling of materials and a particular interest in measurements which would allow materials to be released onto the open market without further control. This section of the report discusses the various techniques which are available for the measurement of specific activity, ie, activity per unit mass. This differs from the other section of the report which concerned itself with the measurement of surface activity, ie, activity per unit area, and is appropriate when either neutron activation has taken place or where contamination has diffused into or become mixed with the material in question.

There is no such thing as a universal technique, given the wide range of nuclides, emission types, emission energies and types and shapes of materials. It seems likely that for any practical, large scale decommissioning of a plant which is in any way complicated then all the methods may have to be employed. It is also inevitable that new techniques and instruments will be developed which will aid the decommissioning process.

7.2 Monitoring of specific α and β activity using hand held equipment

Circumstances may arise where materials have to be monitored which are potentially contaminated mainly by α and β emitting nuclides which have negligible x and γ emissions. Such radiations are difficult to detect in bulk samples. This is ameliorated to some extent, as the proposed clearance levels for the β emitters are much higher than those proposed for the γ emitters. However the levels for the pure α emitters have been set at 1 Bq g^{-1} , the same value as the γ emitters.

The very limited range of α and β emissions dictates that only the surface or near surface of a sample contributes to the signal. Hence it will be essential that either any measurement sample is homogenous or sufficient samples are taken from an inhomogeneous material to allow a confident calculation of the specific activity of a bulk sample equal to the permitted averaging mass.

Problems will also arise for materials which have significant natural activity, such as granite and many soils.

The instruments of choice for work of this nature are large area scintillation detectors or proportional counters filled with a low atomic number gas. The last type can be particularly useful as they are normally capable of measurement of α radiations and also β radiations with maximum energies above approximately 100 keV. Dual α β scintillators have a much higher minimum useful β energy threshold.

The samples should be prepared with as close to a flat surface as possible so that the detector can be supported not more than 2 or 3 mm above the surface. The instrument should then be set to integrate for a suitable period to allow reliable detection of significant contamination. This will depend on the instrument sensitivity, which should be measured for each type to be employed for a representative α nuclide and also for a range of β emitters.

The instrument sensitivity can be determined using suitable sources and a corresponding range of absorbers. For penetrating β emitters, such as $^{90}\text{Sr} + ^{90}\text{Y}$, aluminium sheet 0.5 mm is suitable while for low energy β emitters such as ^{14}C and for α emitters plastic sheet of 1 mg cm^{-2} mass per unit area is suitable. The instrument should be placed on a source of known emission rate per unit area (A) and the net count rate noted (N_1). A single layer of absorber should be placed over the surface, the instrument replaced and the count rate noted (N_2). This process should be continued until the indication is close to background. Sources to be used for this process should have dimensions greater than the detector window and should generate net count rates with no absorber of at least 10 times background to minimise the time required to make the measurements.

The response to distributed activity can be calculated using the following routine.

$$\begin{aligned} \text{Surface emission} &= A \text{ particles cm}^{-2} \text{ s}^{-1} \\ \text{Effective activity} &= 2A \text{ Bq cm}^{-2} \\ &\text{(for 100\% emission)} \end{aligned}$$

Synthesised net count rate from an infinitely thick sample equals

$$\sum_{i=1}^k N_i = N_T$$

where k is the last point above background.

$$\text{Effective activity per unit area} = 2 A \text{ kBq cm}^{-2}$$

$$\text{Effective mass per unit area} = (k-1)P$$

where P is the mass per unit area of each absorber.

$$\text{Instrument response} = \frac{N_T (k-1) P}{2Ak} \text{ s}^{-1} \text{ Bq}^{-1} \text{ g}$$

(for a 100% emission)

This was performed for a Berthold LB1210C and a function derived from the response. The result was

$$R = 2.8 (E - 0.1)^{1.25}$$

where R is the response ($\text{s}^{-1} \text{ Bq}^{-1} \text{ g}$)

E is the maximum energy (MeV)

The value of 0.1 was subtracted from E to generate an effective energy threshold of 100 keV.

As examples, for ^{14}C , the response was $0.1 \text{ s}^{-1} \text{ Bq}^{-1} \text{ g}$ and for ^{36}Cl was $1.7 \text{ s}^{-1} \text{ Bq}^{-1} \text{ g}$. Normal background count rate was 5 s^{-1} , giving a total count over 100 seconds of 500, with a corresponding statistical uncertainty of ± 50 counts at the 95% confidence level or 0.5 s^{-1} . Hence minimum detectable activities for ^{14}C and ^{36}Cl are 5 Bq g^{-1} and 0.3 Bq g^{-1} .

It is also possible to make a fairly crude estimate of maximum energy by covering the sample with one or more of the absorbers used in the calibration exercise. Low energy nuclides, such as ^{14}C , will be attenuated very quickly in comparison with higher energy nuclides.

Low probability γ emissions will also contribute given that they will be relatively unattenuated in their passage through the sample. However the detectors suggested have been chosen for their low x, γ detection probability, which is less than 0.5% for normally incident photons. Hence a sample 1 cm thick, with a mass per unit area of 2 g cm^{-2} , would generate approximately 1 s^{-1} in a 100 cm^2 counter for a level of 1 Bq g^{-1} with a 100% γ emission probability. For low probability ($< 10\%$) γ emissions the β signal will normally exceed the γ contribution.

The same determination of instrument response was repeated for α emitters using a ^{238}Pu source. The corresponding result was $0.3 \text{ s}^{-1} \text{ Bq}^{-1} \text{ g}$. Over a 100 second period such a monitor has a typical background of one or two counts. Hence a level of 1 Bq g^{-1} , generating 30 counts over 100 seconds, would be clearly distinguishable from background and have a random uncertainty (σ) of approximately 40%. Reference 10 illustrates the use of this technique for practical measurements.

7.3 Monitoring of specific γ activity using hand held equipment

Circumstances may arise where it is possible to use conventional radiation protection instruments to estimate the gamma activity per unit mass of samples of decommissioning waste. This technique can be particularly useful for the clearance of large volumes of waste where contamination is unlikely and where the potential contaminant or contaminants are known and are energetic gamma emitters. The technique is not appropriate for samples of complicated shape, wide and varying nuclide mix and for nuclides which emit non penetrating radiations. Typical suitable situations include soil which might possibly have been contaminated by ^{137}Cs from fuel pond leaks, concrete which might contain ^{60}Co activated steel reinforcing or potentially activated steel beams or piping which can be assumed to be uncontaminated.

There are two essentially simple situations. One is where the mass of the material is so large that the sample is close to semi-infinite source and the other is where the sample has a relatively simple shape and where self absorption is very small.

7.3.1 A close to infinite sample

Samples are close to semi-infinite where doubling the mass adds only a few percent to the radiation intensity at the position of the detector. A good example is an excavator bucket holding one or two hundred kg of soil or crushed concrete.

If we consider a hemisphere radius R with the detector placed in contact with the centre of the flat face it is relatively easy to determine whether this represents a close to semi-infinite source. Consider a sphere with the detector at the centre and consider a thin shell at radius r thickness dr. If we assume the detector is spherical and has unit cross sectional area and that the material emits one photon s⁻¹ per unit volume then the fluence contribution for a shell hanging in free space, dφ, is

$$d\phi = 4\pi r^2 dr \frac{1}{4\pi r^2} = dr$$

The first term is the volume considered and the second term is the fractional solid angle subtended by the detector.

However if we assume that the sphere is solid, ie, attenuating, then the contribution, dφ, is given by

$$d\phi = 4\pi r^2 dr \frac{1}{4\pi r^2} e^{-\mu r}$$

where μ is the effective linear attenuation coefficient.

$$d\phi = e^{-\mu r} dr$$

$$\begin{aligned} \phi &= \int_{r=0}^{r=R} \frac{e^{-\mu r}}{\mu} \\ &= \frac{1}{\mu} [1 - e^{-\mu r}] \end{aligned}$$

As R tends to infinity then e^{-μr} tends to zero.

Hence if the dimension of the source is such that e^{-μr} is less than 0.2 then the source can be taken as acceptably close to infinite.

Solid concrete, as an example, has a density of 2.7 g cm⁻³. The effective attenuation coefficients for ¹³⁷Cs and ⁶⁰Co⁽¹¹⁾ are given in Table 22 as are the radii in cm corresponding to e^{-μr} = 0.2.

TABLE 22 Thickness of concrete representing dose to an infinite source

Nuclide	Attenuation coefficient (μ)	R for e ^{-μR} = 0.2 (cm)
¹³⁷ Cs	0.10	16
⁶⁰ Co	0.08	20

A hemisphere of this dimension would have a mass of 86 kg, which is much less than the mass held in most excavator buckets.

The dose rate at the centre of an infinite mass can be calculated simply from the gamma energy per disintegration and the activity per unit mass. Any beta decay should be ignored as the majority of equipment used in assessing waste activity is normally insensitive to beta radiation for reasons of robustness and also because variations in activity distribution produce large variations in beta dose rate.

$$\text{Hence, the dose rate, } \underline{D}, \text{ in } \mu\text{Gy h}^{-1} = A \times \Sigma E \times 1.6 \times 10^{-13} \times 10^6 \times 3600$$

A = activity concentration (Bq kg⁻¹)

ΣE = total γ energy emitted per decay (MeV)

1.6 x 10⁻¹³ converts MeV to joules

10⁶ converts Gy to μGy

3600 converts Bq to disintegrations per hour.

$$\text{Thus } \underline{D} = A \times \Sigma E \times 5.8 \times 10^{-4}$$

For ⁶⁰Co ΣE = 2.5 MeV

Choosing A = 1 kBq kg⁻¹ gives a dose rate of 1.44 μGy h⁻¹ within an infinite mass.

At the surface of a semi-infinite mass the dose rate to the material will be something less than half that at the centre of an infinite mass. It will not be exactly half as the backscatter from the solid material will not be reproduced if the interface is with air. The dose rate to air will also differ slightly from the dose rate in the material because of differences in the mass energy attenuation coefficients. Hence, it is possible to make a reasonably good, but not perfect, calculation of the dose rate to air in air at the surface of an effectively semi-infinite contaminate medium. Following this argument through gives a value of air dose rate for a semi-infinite source contaminated by ⁶⁰Co at a level of 1 kBq kg⁻¹ of approximately 0.7 μGy h⁻¹. Such a level is not normally easy to measure with instruments designed for the measurement of occupational dose rates where there is generally a trade off between sensitivity and the achievement of good dosimetric characteristics, such as a flat energy response and good polar response. However, there is a large number of sodium iodide based instruments which will detect such a level with ease. As an example a 50 mm x 50 mm sodium iodide scintillation detector will have a response of approximately 700 s⁻¹ μGy⁻¹ h for ⁶⁰Co γ radiation when used with a simple, ie, non energy selective, ratemeter. The background on such an instrument is usually in the range 20 to 50 s⁻¹. A level of 0.7 μGy h⁻¹ ⁶⁰Co would then generate an indication of approximately 500 s⁻¹ above a background of 50 s⁻¹. For ¹³⁷Cs 1 kBq kg⁻¹ will generate a surface dose rate of approximately 0.2 μGy h⁻¹ and with the detector's enhanced response of approximately 1000 s⁻¹ μGy⁻¹ h will give an indication of 200 s⁻¹ above background. The actual values will be higher than this because of the rising response of the sodium iodide scintillation detector at low energies. The gamma radiation spectrum at the surface will comprise a mixture of primary γ-rays of the expected energy and scattered photons of lower energy. The detector will have a higher response to those energies when compared to the primary energy.

It is also essential to establish a reference background for the measurement, which should replicate as closely as possible the circumstances of the operational measurement using the same position and sample container, with the only difference being the use of a clean sample of the material to be assessed. For many materials the background indication of the instrument when it is placed, for

example, at the centre of the top surface of a loaded excavator bucket, will be lower than the free air value, because the bucket and contents shield the detector from a large proportion of the terrestrial gamma radiation. However, with materials such as bricks which have relatively high levels of natural radionuclides, then the background will increase.

The technique as described has a relatively high level of uncertainty. The source is not truly semi-infinite, which will lead to an underestimate of the activity concentration. The radiation spectrum can have a relatively large scatter component, which will normally lead to an overestimate of the activity concentration. The latter point can be addressed by using a gated ratemeter. For a 50 mm x 50 mm detector the photo fractions for ^{137}Cs and ^{60}Co are typically approximately 0.39 and 0.23 for a well set up instrument according to manufacturer's data ie, approximately 39% and 23% of the interactions will generate a photoelectron and hence a full energy peak. This would reduce the instrument responses to $400 \text{ s}^{-1} \mu\text{Gy}^{-1} \text{ h}$ for ^{137}Cs and $160 \text{ s}^{-1} \mu\text{Gy}^{-1} \text{ h}$ for ^{60}Co . For sample levels of 1 kBq kg^{-1} this would give count rates of 80 s^{-1} (^{137}Cs) and 110 s^{-1} (^{60}Co), but against a background of a few counts per second rather than 50. Hence, it is possible to reduce the uncertainty due to scattered radiation.

For optimum results, and to generate a very obvious and robust traceability to national standards, the best technique is to generate a uniform, well mixed, sample, perform the measurements using the survey meter, take a series of samples from the bulk sample and then count these using an intrinsic germanium semi-conductor detector in a well defined volume for which the detector efficiency is well known.

The well mixed sample can also be used to check the variation in response of the instrument over the top surface of the sample and also to generate correction factors for measurements through the side of the excavator bucket. It will then be possible to make measurements over the whole sample container which can be useful if the sample is not close to homogenous, ie, there is the possibility of localised hot spots which could compromise the disposal process. A typical process would involve a grid of measurement points spaced at intervals equal to approximately half the smallest dimension (d) of the bucket. If a localised hot spot is present then it would be unlikely to be more than $\frac{1}{2}d$ away if the bucket is rectangular. A measurement routine can be developed whereby the mean corrected activity is assessed using the results from all the monitoring points and where statistically significant deviations from the mean are assumed to be caused by an individual hot particle.

The simplest method of determining this is to place a relatively active source of the relevant nuclide within the sample at the appropriate position and then noting the instrument count rate. This may not be practicable and a less effective but still useful method may have to be employed. This uses a calculation of the expected transmission of the sample and a measurement of the instrument response to a point source. This should be performed by mounting a source of $k \text{ Bq}$ in free air at a distance of $x \text{ cm}$ from the detector centre and then determining the net count rate n . The response is then $n k^{-1} x^2$. This value can be used to determine the instrument response to a source at the least favourable distance, $y \text{ cm}$, which will be

$$\frac{n k^{-1} x^2}{y^2} \text{ counts s}^{-1} \text{ Bq}^{-1}$$

If the expected transmission is T at $y \text{ cm}$ then the final instrument response is:

$$\text{response (s}^{-1} \text{ Bq}^{-1} \text{ at ycm)} = \frac{n k^{-1} x^2 T}{y^2}$$

Again for a 50 mm x 50 mm sodium iodide detector and a depth of 30 cm in crushed concrete the response will be approximately $1.0 \text{ s}^{-1} \text{ kBq}^{-1}$ for ^{60}Co , giving a minimum detectable activity, corresponding to a doubling of background, of 50 kBq , ^{60}Co .

These techniques can also be employed where there is a range of nuclides present provided that the ratios of the activity concentrations are not expected to vary greatly or where the waste sentencing procedure is dominated by the penetrating gamma emitters and the other nuclides are, pro-rata, much less important.

The simplicity of this technique allows it to be used effectively by a relatively unskilled work force employing simple and robust equipment. Measurements can be performed where the material is being removed. It is particularly suitable where it is necessary to check large volumes of material which are expected to be essentially uncontaminated or where the expected level is well below the limit set. The main difficulties may well occur when monitoring materials with relatively high levels of natural radioactivity. These will give much enhanced background count rate in both the simple and gated rate meter modes, and thus raise the limits of detection.

Another aspect which can be a problem with gated systems is the variation of gain with temperature of a sodium iodide scintillator - photo multiplier tube combination. A typical gain change over the temperature range 0°C to $+40^\circ\text{C}$ is $\pm 5\%$ compared to $+20^\circ\text{C}$. Some systems incorporate gain stabilisation but otherwise this temperature dependency demands either regular adjustment of the instrument window settings or high voltage, which is extremely undesirable, or that the upper and lower thresholds be set so that the photopeak always lies within them over the expected temperature range. This approach increases the background count rate significantly.

7.3.2 Samples of limited thickness

Some samples such as tubes can be assessed very simply. If a thin walled metal tube is considered then it is possible to make a series of simplifying assumptions if it is being monitored for the presence of energetic gamma activity. These are that absorption within the sample is close to zero and that a side shielded detector mounted in contact with the surface will essentially see radiation equivalent to that emitted from the surface directly beneath it. The reasoning behind this is the approximation that in a long thin tube all the radiation escapes and its intensity is uniformly distributed over its area. In practice the sides of a detector will contribute some signal and hence any measurement will tend to overestimate the activity level. However this effect can be reduced by side shielding the detector.

For samples of significant thickness correction can be made for attenuation through the sample, either by calculation or, in some cases, by performing a measurement using the relevant source.

The measurement of activity per unit mass is less straightforward when dealing with thin walled tubes or sheets when compared to one measurement of bulk activity, simply because the available mass of sample is much reduced. Consider a thin walled tube, density ρ , wall thickness x mm, diameter y mm and neutron activated, giving a ^{60}Co level of $m \text{ Bq kg}^{-1}$. What response would we get using a sodium iodide scintillation detector with an overall detection efficiency of 70% for normally incident ^{60}Co photons and area $A \text{ cm}^2$?

Calculate the emission rate per unit area of pipe; -

$$\begin{aligned} \text{thickness} &= X \text{ mm} \\ \text{density} &= \rho \\ \text{activity} &= m \text{ Bq kg}^{-1} \end{aligned}$$

Thus 1 cm^2 will contain

$$\begin{aligned} &\frac{X}{10} \cdot \rho \cdot \frac{m}{1000} \text{ Bq} \\ &= 10^{-4} X \rho m \text{ Bq} \end{aligned}$$

The area of the detector is $A \text{ cm}^2$ and the detection efficiency is 70%.

As the contaminant is ^{60}Co , which emits two photons per disintegration, then the count rate will be

$$\begin{aligned} &= 10^{-4} X \rho m \times A \times 2 \times 0.7 \\ &= 1.4 \times 10^{-4} X \rho m A \text{ s}^{-1} \end{aligned}$$

If we insert realistic values

$$\begin{aligned} \text{ie, } X &= 3 \text{ mm} \\ \rho &= 8 \text{ g cm}^{-3} \\ m &= 1000 \text{ Bq kg}^{-1} \\ A &= 20 \text{ cm}^2 \end{aligned}$$

then the count rate

$$\begin{aligned} &= 1.4 \times 10^{-4} \times (3 \times 8 \times 1000 \times 20) \text{ s}^{-1} \\ &= 67 \text{ s}^{-1} \end{aligned}$$

A typical background count rate would be 25 to 50 s^{-1} for such a detector, with a thickness of 50 mm.

Hence a level of 1000 Bq kg^{-1} will effectively double the count rate.

For ^{137}Cs the corresponding value would be approximately 35 s^{-1} for 1000 Bq kg^{-1} .

Such levels are measurable using a rate meter with a typical time constant. However it will often be useful to make a measurement of the level, rather than indicate only that it is acceptable for recycling. This can be done with many instruments using the integration mode. With this technique the integration period is started and the detector moved over the surface of the object to be monitored. With care it will be found feasible to monitor such objects as 2 m long thin walled pipe in such a way that the detector has been scanned fairly evenly over a large part of the area of the object. This would give the following results.

Counting time = 100 s

Typical background count = $5 \times 10^3 = \text{BG}_L$

(This can be established with a high level of precision by making several measurements on a known clean sample of the material).

As the background count rate is well known then the statistical uncertainty is dominated by the measurement of the real sample, ie, the count generated by source, S, and background, BG.

$$\begin{aligned} \text{Net count} &= (S + BG) - (BG_L) \\ &\approx S \\ \text{Uncertainty } (\sigma) &= (S + BG)^{1/2} \end{aligned}$$

The minimum level of detection (at the 95% confidence level) will be when

$$S = 2 (S + BG)^{1/2}$$

using $BG = 5000$ gives a value of $S = 140$ or 1.4 s^{-1} which corresponds to minimum detectable activities of $0.02 \text{ Bq g}^{-1} \text{ }^{60}\text{Co}$ and $0.04 \text{ Bq g}^{-1} \text{ }^{137}\text{Cs}$.

Hence it should be possible to identify clean samples as having levels of one tenth or less of the appropriate limit.

In areas with abnormally high background levels, advantage can be gained by using a more sophisticated scaler rate meter equipped with one or more counting channels, provided the likely contaminants are well known. The most obvious circumstance where this technique shows advantage is where the monitoring area is irradiated by scattered radiation produced by the nuclides to be measured. In such a case there can be a relatively high intensity of scattered radiation, which is of a lower energy than the primary, with only a relatively small component of the primary radiation. Using a counting channel centred on the photopeak of ^{60}Co will give a low background count rate allowing the minimum detectable activity to remain unchanged. For example, for a 50 mm x 50 mm detector, for ^{60}Co .

$$\begin{aligned} \text{Total detection probability} &= 60\% \quad \text{for normally} \\ \text{Photopeak probability} &= 20\% \quad \text{incident photons} \\ \text{Photopeak count rate generated by } 1 \text{ Bq g}^{-1} \text{ }^{60}\text{Co} &\approx 20 \text{ s}^{-1} \\ \text{Typical background} &= 5 \text{ s}^{-1} \end{aligned}$$

Hence threshold of detection (allowing for a well established background) and a counting time of 100 s will be 0.02 Bq g^{-1} . It may seem strange that no advantage has been gained from gating the unit but this is because of the relatively low photopeak detection probability of an easily portable detector. However, it has remained unaffected by the higher background.

A problem arises where there is a mix of nuclides, where Compton scattered radiation from a high energy emitter can appear in the photopeak for a lower energy emitter, thus increasing the uncertainty. This will be compensated for, to some extent, by the rapidly rising photopeak efficiency of sodium iodide scintillators. As an illustration the photopeak efficiency for ^{137}Cs is 1.8 times that for ^{60}Co for a 50 mm thick detector.

In summary it is possible to sort samples of limited thickness such as thin walled tubes and plate into three groups; the excessively contaminated, the clearly acceptable and, inevitably, given the uncertainties, those samples that lie on either side of the acceptable limit. However the technique is simple, suitable for use by unskilled workers and requires no fixed monitoring stations. Uncertainties

can be reduced by taking samples, counting them using more sophisticated equipment and using the resulting values to generate better calibration values for the hand held equipment.

7.3.3 Thicker samples and samples of complicated shape

Two sample approximations have been discussed, one where the sample is close to infinitely thick and the other where the sample is very thin, in terms of the mean free paths of the gamma radiations. For other configurations the situation is much more complicated, ie, thick samples where there is still a significant probability of a photon emitted from the far surface of the sample reaching the detector, and samples where the shape is complicated, so that the assumption that the radiation field is fairly uniform over the surface no longer holds.

One particularly obvious and common example of both situations is the steel "I beam", used in building structures. Typical sizes are depths up to 0.5 m and web thicknesses of 20 mm. Steel with a thickness of 20 mm has a transmission of approximately 60% for ^{60}Co γ radiations and 50% for ^{137}Cs (11). Monitoring of such beams can be carried out using modified portable equipment. The detector should be equipped with a collimator which restricts the influence of radiation incident from the sides. The limiting factor on the detector size is the maximum weight which a user could be expected to manipulate in construction site conditions. For example a thickness of 37 mm of lead has a transmission of 20% for ^{60}Co (11). If a collar of that thickness was placed round a 50 mm detector then it would have a mass of 1.1 kg per cm of length. It would then just be possible to use a 50 mm x 50 mm detector with 37 mm shielding around the scintillation crystal with a gradual stepping down of shielding thickness around the photomultiplier tube, leading to a total detector weight of approximately 12 kg. Such shielding would essentially limit the detector's response to the area immediately below the window.

Compensation for the thickness of the sample can be performed either by calculation or by actual measurement. The latter is more reliable but it may be administratively unacceptable, because of the risk of loss of sources under site conditions, and it may also be difficult to place sources reliably inside such objects as valve cavities. In such cases it may be impossible to perform on site measurements effectively without cutting the object up into more convenient shapes, with the attendant potential for spread of contamination and dose uptake.

7.3.4 Hand held equipment with spectrometer capability

Developments in signal processing power have lead several manufacturers to produce scintillation detector based instruments with spectrometric capability. These vary from completely self contained instruments which will acquire spectra and identify contributions from up to 10 nuclides, to instruments which acquire and store spectra, and require connection to a personal computer for interpretation. Such instruments are inevitably more expensive and will take longer to make each measurement, because of the processing involved. However they are extremely useful in checking that the expected mix of gamma radionuclides has not changed dramatically and that the calculated maximum acceptable count rates used for the simpler equipment are still valid.

7.3.5 Calibration of gamma monitoring equipment

Sodium iodide scintillation detectors and other similar devices are simple to set up and calibrate if they are to be used with only a low energy counting threshold. Their energy detection probabilities are also well documented. Calibration can be by two different routes. One uses distant dosimetric sources in a facility designed for the calibration of conventional radiation protection instruments. The other uses point sources of much lower activity designed for the energy calibration of spectrometers.

The aim is to demonstrate that

- (1) the energy threshold is acceptably low;
- (2) the detection efficiencies correspond to the expected values for the detector size;
- (3) the response is linear, ie, doubling the number of incident photons of a defined energy doubles the net count rate;
- (4) the background conforms to the expected value.

7.3.5.1 Using dosimetric sources

For most purposes an energy threshold of approximately 40 keV will be suitable. This gives a suitable factor above system noise and there is no real point going to a lower threshold as the attenuation of gamma radiation rises very rapidly for most materials below 60 keV. This can be achieved using a dosimetric ^{241}Am γ source, ie, one designed to generate 60 keV photons only, and increasing the detector supply voltage until it is just on a plateau. For typical detector sizes of 50 mm x 50 mm and typical source activities of 0.4 GBq a very large source to detector distance will be required, of approximately 10 metres, to avoid generating very high count rates. For many of these sources the measured quantity will be air kerma rate in $\mu\text{Gy h}^{-1}$. This can be converted to photons $\text{cm}^{-2} \text{s}^{-1}$ by dividing the value in $\mu\text{Gy h}^{-1}$ by the appropriate value given in ICRP publication 74⁽¹²⁾ for the air kerma per unit photon fluence. Even at distances of 10 metres the radiation intensity in a well designed facility should obey the inverse square law as the air absorption is low, $\approx 1\% \text{ m}^{-1}$ and hence it should be sufficiently accurate to calculate the dose rate at, for example, 10 m based on the most distant measured dose rate. The detection efficiency for a good crystal should be very close to 100% for ^{241}Am photons.

The instrument should then be exposed to collimated beams of ^{137}Cs and/or ^{60}Co γ radiations and the expected detection efficiencies confirmed. Again, using data from ICRP publication 74⁽¹²⁾, $1 \mu\text{Gy h}^{-1}$ air kerma is equivalent to 80 and 54 photons $\text{cm}^{-2} \text{s}^{-1}$ for ^{137}Cs and ^{60}Co γ radiations respectively. The measured responses should confirm to the expected values. A suitable reference is Figure 10.22 of Knoll (see Bibliography). The measurement should be repeated at another dose rate to confirm the linearity of the instrument. Dose rates of less than $1 \mu\text{Gy h}^{-1}$ are required for 50 mm x 50 mm detectors. The background count rate should also be determined.

If a measurement of total (gamma + cosmic) dose rate is available this should be multiplied by the ^{60}Co or ^{137}Cs response factors for the scintillation detector and the results compared with the measured background count rate. Agreement within a factor of 2 is to be expected.

7.3.5.2 Using energy calibration sources

Calibration using these sources is rather simpler, simply because of their relative ease of manipulation.

Again the threshold should be set in the region of 40 keV. There is an additional complication in the use of a ^{241}Am point source in that the γ emissions at 26 keV and the x ray lines in the 13 to 20 keV region will also be present⁽⁷⁾, unlike in the dosimetric source where the greater thickness and

encapsulation will normally virtually eliminate them. Care should be taken to ensure that the first count plateau is used, as the high voltage is increased. The detection efficiency should be close to 100% in terms of the probability of detecting a 60 keV photon striking the detector.

Normally if the threshold is set at a much lower energy the background count rate will be higher than expected.

Measurements should also be made with ^{137}Cs and ^{60}Co point sources and the results compared with the expected values. Note that there may be a contribution of a few per cent from the K x ray lines from ^{137}Cs in the 31 to 36 keV range, a small proportion of which will be counted if the effective threshold is close to 40 keV.

Measurement of linearity can be confirmed by making measurements at two distances. Very close approach to the detector should be avoided as the detection probability will rise when the detector subtends a large solid angle because of the increase in mean path through the crystal.

For instruments with gated systems it will be essential to set up each window to correspond to the photopeak of the radiation of interest. Portable equipment may well not have good energy linearity, ie, the photopeak for ^{137}Cs (662 keV) should correspond to a pulse amplitude approximately 11 times that of ^{241}Am (60 keV) but it may, in practice, be much lower. There is a balance between energy linearity and power consumption which makes it difficult for manufacturers to offer both spectrometric standards of energy linearity and good battery life.

7.4 Problems associated with the use of scintillation detectors and photomultiplier tubes

The two main problems are damage to the crystal and the influence of magnetic fields. Sodium iodide is a relatively brittle material with a strong hygroscopic tendencies. Rapid changes in temperature and mechanical shock can lead to fracture of the crystal, effectively reducing the detector volume and the light detection efficiency. Holes in the can around the crystal will allow the entry of water vapour, generating areas of hygroscopic degradation, in the shape of pockets of yellow powder. These lead mainly to a reduction in collection efficiency causing a change in effective threshold. It is important to protect these detectors as well as possible and to arrange regular checking of performance using ^{241}Am γ radiation (60 keV).

Magnetic fields can also influence the effective threshold of an instrument, by defocusing the electron beam in the photomultiplier tube which reduces the gain. Well designed instruments have a mu metal shield around the detector, which acts as a magnetic short circuit, but inevitably protection is not perfect and the point will be reached where the performance of the instrument is affected. Monitoring close to surfaces of objects that are intentionally magnetic will always be a problem. However it is also common for large steel beams to have significant magnetic fields associated with their manufacture. If there is any doubt that the instrument's performance has been affected then it should be checked by fixing a small ^{241}Am source close to the detector. The count rate due to the source should not change significantly when the detector is moved over the object to be monitored. This shows that the effective energy threshold is still less than 60 keV. If it does then the results of monitoring may be in doubt.

If strong magnetic fields are likely to be a problem then a possible solution is the use of a caesium iodide scintillator coupled to a large area PIN diode. The combination has a higher effective noise level but this will not drastically affect counting efficiencies for high energy gamma emitters. Caesium iodide is also a much tougher material, resistant to fracture, and is much less hygroscopic.

7.5 Monitoring of specific activity using installed equipment

Monitoring using installed equipment offers both advantages and disadvantages over monitoring with portable equipment. The advantages are that detectors can be larger, a wider range of detector types can be employed, shielding and collimation is much easier and the available electronic processing power is much greater. The disadvantages are that the equipment is much more expensive and that the objects to be monitored have to be transported to the monitoring station. Inevitably, then, some in situ monitoring will have to be performed on materials before they are consigned.

7.5.1 Conveyor belt monitoring using sodium iodide scintillation detectors

Conveyor belt monitoring is very attractive when large quantities of waste are to be processed which can easily be broken up into relatively small pieces. In such cases the detectors can be large pieces of sodium iodide scintillator, for example, 75 mm x 125 mm x 400 mm. One detector is normally placed above the belt across the line of motion of conveyor belt and one beneath it. This arrangement minimises the effects of self attenuation within the material and also reduces the influence of height variations in the sample dramatically, as the increase in solid angle subtended by the upper detector produced by a vertical displacement off the conveyor belt will be virtually balanced out by the reduction for the lower detector. The detector can be extremely well shielded with the additional effect that the back shielding for one detector also acts as a shadow shield for the opposing detector.

The detectors are then normally connected to a multichannel analyser. This provides information on the artificial activity within the sample but also provides a much larger degree of rejection of signals produced by naturally radioactive elements. A very important example is the 609 keV line from ^{214}Bi which is relatively close to the 662 keV line from ^{137}Cs . Note that scintillation detectors have relatively poor energy resolutions, typically of approximately 10% full width at half maximum, which limits their ability to discriminate between energies, especially at low total counts from each energy.

The detectors can be temperature stabilised to a high degree. This is normally performed by placing a small source within the detector envelope, often ^{241}Am used as a gamma emitter. The high voltage applied to the detector is controlled to give a particular ratio of counts between two adjacent counting windows. The only problem with the technique is that the presence of the stabilisation source reduces the detection ability for that nuclide dramatically.

The loading depth and conveyor speed is chosen to comply with any restriction on averaging mass and also to produce acceptable uncertainties caused by self absorption within the sample. It is possible to generate direct data on sample absorption by placing a relatively active source on one side of the belt with a detector on the other between the conveyor loading hopper and the activity monitoring station. The output from this unit can be used to correct the apparent measured contamination level. The load on the conveyor can also be measured in mass per unit length which can also be correlated with the monitoring results to give a relatively accurate activity concentration.

The measurement process on the conveyor is a continuous one, unlike many of the other measurements described in this report where the detector is placed on a sample and its indication noted. The measurement integration time should be chosen to correspond with the time for which a particular part of the sample is in view as it passes along the conveyor under the collimator. This ensures that a particular measurement corresponds to an identifiable volume of sample.

With the detectors suggested it will be possible to measure ^{137}Cs at a level of 0.1 Bq g^{-1} in low background activity material using a 5 second monitoring period, and a collimator viewing approximately 30 cm of the conveyor belt length. Conveyor belt speed would then be approximately

3 metres per minute, giving a throughput of the order of 300 kg per minute. Similar results would apply for other energetic γ emitters, after making allowance for decay probabilities.

The conveyor belt can be controlled to sort the material into 3 volumes; the definitely acceptable, material on either side of the acceptable level and the unacceptably contaminated. The centre volume could be run through again at a slower speed, which would increase the precision of the measurement and reduce the volume of material which is not acceptable for reuse.

7.5.2 Conveyor belt systems using plastic scintillators

In some circumstances large volumes of waste need to be processed where it is expected that there is no actual contamination or activation. For materials which have a low inherent background level, such as some concrete, then a significant saving in money and complexity can be produced by using very large plastic scintillators as the detectors. These scintillators are quite efficient, in the sense that the probability of a gamma photon passing through the detector without interacting is low. For example for a 10 cm path length in plastic the probability of ^{137}Cs and ^{60}Co photons escaping without detection are 40% and 50% respectively. However spectrometry is not feasible because it is unlikely that the full energy of an energetic photon will be deposited. Photoelectric interactions are unlikely, and the attenuating properties of the materials do not rise rapidly as the photon energy decreases, unlike sodium iodide and germanium. Hence Compton scattered photons may well escape from the detector, either directly or after another interaction.

Scintillator volumes of 10s of litres are possible, giving a factor of increase over a typical large sodium iodide detector of a factor of 10. The material is much tougher and is not susceptible to water damage. It is also much more resistant to rapid temperature changes.

In a conveyor belt system, with relatively low background material, then energetic γ emitters can be detected at levels of 0.1 Bq g^{-1} at the same conveyor belt speed etc used earlier.

7.6 Monitoring for γ emissions using semiconductor detectors

7.6.1 The properties of semiconductor detectors

The method of choice for situations where a wide range of nuclides may be present is the intrinsic germanium semiconductor spectrometer. These devices have the ability to determine the energy of gamma emissions with great accuracy and to separate effectively emissions of very similar energy. In this way they are superior to sodium iodide based equipment, which has a much poorer energy resolution.

They are inferior to sodium iodide and other scintillation equipment in some ways. It is not possible to produce single detectors of very great size. The atomic number of germanium is lower than that of iodine leading to a lower probability of photoelectric interaction. Current technology can offer crystals which have 1.5 times the ^{60}Co photopeak detection efficiency of a 76 mm x 76 mm sodium iodide detector. (Sodium iodide crystals are routinely available with 20 times that efficiency.) They are also more expensive, require a more sophisticated preamplifier and have to be operated at low temperatures. This requires either an electrical refrigerator or a supply of liquid nitrogen. The cooling system inevitably adds bulk to the system and the detectors are also susceptible to vacuum leaks, leading to a loss of insulation and a rise in liquid nitrogen consumption.

Inevitably, then, the decision on which type of spectrometric equipment to employ requires the balancing of the metrological advantage of the intrinsic germanium detector with the lower cost and greater convenience of the scintillation detector.

Knoll (see Bibliography) gives a comprehensive description of the physics of these detectors, and the various factors which should be taken into consideration when selecting a detector. Given the importance of these detectors, however, the main conclusions are summarised here.

7.6.1.1 *Energy resolution*

The energy resolution of a good germanium based system designed for high energy gamma monitoring is typically 2 keV at 1.33 MeV, one of the ^{60}Co lines, ie, approximately 0.2%. This can be compared with a typical value for sodium iodide of 7%. Both these values approach the theoretical limit, derived from the number of holes and electrons generated in the germanium crystal and the number of photoelectrons generated in the photomultiplier tube connected to the scintillator.

7.6.1.2 *Peak to Compton ratio*

Germanium has a relatively low atomic number (32) and hence the probability of a photoelectric interaction is correspondingly lower than that for iodine (53). Events corresponding to a full energy peak are much more likely to be based on multiple Compton scattering followed by photoelectric absorption of the scattered photon. However, for comparable sizes or efficiencies a greater fraction of all detected event will be in the Compton continuum. The improved spectrometric characteristics do give rise to a much more faithful reproduction of the Compton edge. Peak to Compton ratios are often quoted for detectors. This is defined as the ratio of the count in the highest photopeak channel to the count in a typical channel of the Compton continuum associated with that peak. Normally the 1333 keV line from ^{60}Co is compared with the average count in the 1040 to 1096 keV region. Typical values range from 30 to 60. This high value illustrates that the very high energy resolution more than compensates in many circumstances for the relatively low photopeak detection probability.

7.6.1.3 *Escape peaks*

Escape peaks play a much more important role in germanium detectors than sodium iodide detectors because of the much higher probability that a germanium characteristic x ray may escape from the detector. This can lead to a small peak 11 keV below the main peak, and is particularly obvious for low energy lines.

The effect is even more prominent for very high energy interactions, where pair production generates a positron which can combine with an electron, generating two 511 keV photons. One or both of these can escape, leading to two peaks 511 keV and 1022 keV below the main peak.

In addition 511 keV photons from outside may also be detected, giving a prominent 511 keV peak. Such high energies are not normally a problem in decommissioning because of the short half life of high energy γ emitters. However the decay products of ^{226}Ra , which can be an important component of any natural activity present, have energies up to 2.45 MeV, and will generate these effects.

7.6.1.4 *Prediction of response*

Sodium iodide detectors are generally manufactured in a limited range of sizes, the characteristics of which are well documented. Intrinsic germanium detectors are made in a wider range of shapes and sizes and the most effective way of determining a particular detector's characteristics is normally by using a Monte Carlo based code, eg, MCNP. A limited series of measurements is used, essentially, as input data leading to a prediction of response over the detector's operating range. This cuts the cost of detector commissioning.

7.6.2 Monitoring applications

7.6.2.1 Finger printing

The merits of the germanium and sodium iodide detectors can frequently be combined by finger printing materials. In this technique samples are taken of the material of interest and assessed using a germanium detector for the γ emitters and any appropriate techniques for the α and β emitters (see section 7.9). This generates a detailed record of the emitters present. The contents can be assessed and a limit chosen for a prominent γ emitter, in Bq g^{-1} , based on the total radioactive content. Sodium iodide based equipment can be used to assess the effective activity of samples by gating it on that γ line and setting the rejection level at an appropriate point. Hence the higher precision, higher cost equipment can be used to control a larger number of simpler, operationally more convenient, units.

A regular sampling programme is required either to demonstrate that the original nuclide mix is being maintained or to correct the setting of the sodium iodide equipment. Changes in nuclide mix can occur, for example, as a result of the change in neutron spectrum with distance from a reactor giving rise to different nuclide ratios in a steel beam.

7.6.2.2 Drum scanner

Drums holding 200 litres are frequently used to hold materials such as soil or metal swarf from a size reduction exercise. A typical use for germanium detector is in a drum scanner. This comprises a detector, provided with a collimator, which is mounted close to a drum, mounted on a turntable. The collimator is designed so that it sees, typically, about 10% of the height of the drum. The drum is rotated and spectral data generated for the first segment. The detector is then lifted, so that the detector is viewing the adjacent disc shaped segment of the drum. This process is repeated until the entire drum has been scanned. The segments are analysed in turn and a complete inventory of the contents generated. This technique is thus capable of measuring total activity and of identifying any excessively high activity volumes within the drum.

Self absorption within the drum is important for all barring low atomic number, low density materials. Corrections can be applied based on average waste density, or differential absorption, in which the correction is derived from the difference in apparent activity of a particular nuclide derived from different γ energies emitted by that nuclide, or by a transmission source measurement in which the attenuation of an external source is used to generate a correction.

The performance of such a system depends on the detector size and the number of segments. Larger detectors and a more open collimator will generate either faster answers or a lower limit of detection. For a drum filled with low density material (0.2 g cm^{-3}) one manufacturer quotes a detection limit of:

^{239}Pu	0.2g,	$5 \mu\text{g g}^{-1}$
^{137}Cs	$5 \rightarrow 10 \text{ kBq}$,	$0.13 \rightarrow 0.26 \text{ Bq g}^{-1}$
^{60}Co	$5 \rightarrow 10 \text{ kBq}$,	$0.13 \rightarrow 0.26 \text{ Bq g}^{-1}$

This assumes a 30 minute monitoring time and a 30% efficient detector, and is without any shielding around the drum.

A similar system can be constructed inside a completely shielded enclosure built from low background steel, and employing multiple detectors. Typical performance for a 10 minute counting time and three 20% germanium detectors is given below in Table 23.

TABLE 23 Typical drum scanner detection limits

Nuclide	Lower limit of detection (Bq g ⁻¹)	
	Average density 0.1 g cm ⁻³	Average density 1.8 g cm ⁻³
¹³⁷ Cs	7 x 10 ⁻³	1.5 x 10 ⁻³
¹³⁴ Cs	6 x 10 ⁻³	1.1 x 10 ⁻³
⁶⁰ Co	6 x 10 ⁻³	1.1 x 10 ⁻³
²³² Th	2 x 10 ⁻²	4 x 10 ⁻³
²³⁵ U	1 x 10 ⁻²	4 x 10 ⁻³
²³⁹ Pu	400	100

Similar limits of detection will apply for other energetic γ emitters, after taking account of decay probabilities.

7.7 Calibration of intrinsic germanium detectors

7.7.1 Energy calibration

Even the best detector systems will have a small degree of energy non-linearity. When dealing with waste which is contaminated by a limited range of nuclides which have distinct γ energies then this is not a problem, as the waste acts as its own energy calibration. However when waste materials contain, or may contain, a large number of nuclides with complicated energy spectra and differing free release acceptable levels, then an accurate energy calibration is essential. This should be undertaken with a standard source or sources with energies which cover the range to be assessed in practice. This source should have a reasonably long half life, be obtainable in a pure form, and not be subject to the ingrowth of γ emitting progeny. A popular source is ¹⁵²Eu, which emits 20 different energies, covering the range 122 keV to 1458 keV. With the addition of ²⁴¹Am (59.5 keV) this offers a good energy calibration for the majority of circumstances.

Data from this calibration can be used to correct the apparent energy of the γ emissions from the waste to generate a true energy. The observed peaks can then be analysed, generally using software provided by the detector system manufacturer, to provide an accurate identification of the nuclides present. For monitoring of large volume samples with high self attenuation the relative intensity of γ emissions generated by one nuclide can be used to assess the degree of self absorption within the sample. Software will also normally indicate if any peaks have not been identified and also if there is an apparent inconsistency, where, for example, the activity of ⁶⁰Co derived from the 1.173 MeV peak does not agree with that derived from the 1.332 MeV peak.

Effective energy calibration is thus a useful tool in checking that a system is working correctly.

7.7.2 Efficiency calibration

Efficiency calibration is more complicated as it is influenced by the shape, density and atomic number of the sample. The purpose of the efficiency calibration is to yield a number which can

be used to relate the height of a photopeak above the underlying continuum to a specific activity in a real sample.

Standards are readily available in the form of a small deposit on a very thin backing material. These approximate closely to a point source. For large volume samples, such as water and soil, calibration standards are available for a range of nuclides in the form of a Marinelli beaker. This is a beaker which fits closely over the end cap of the detector. There is a range of standard sizes.

However, the situation for practical samples encountered during decommissioning is generally more complicated. Typical samples include drums, beams and pipes. The best option in these circumstances is to use a detector with known point source efficiency, generally derived using a mixture of measurement and calculation, and a calculated correction factor, based on the size, shape, density and atomic number of the sample. Monte Carlo based programmes, such as MCNP, can be used to derive such correction factors. Generally these are derived for a limited series of sample geometries and compositions and then each sample is either assumed to conform to one of that series or, for complicated shapes, is cut up and packaged into a form which conforms to one of the series. It is this last step which generally generates the largest uncertainty in the monitoring process.

7.8 Assessment of transuranic waste

For the majority of transuranic materials γ monitoring is not adequate to allow the free release of potentially contaminated materials. A notable exception is ^{241}Am but even then the 59.5 keV γ emission is rapidly attenuated by even thin steel sheet or one or two cm of concrete.

Two techniques using neutrons offer some assistance but even these will not generally permit the free release of materials, and hence strictly do not fall within the remit of this document. The simplest technique uses passive neutron counting. This employs an array of ^3He filled proportional counters around the drum. These counters are very efficient detectors of thermal neutrons but the efficiency falls rapidly, approximately as the square root of the energy, as the neutron energy increases. To compensate for this the detectors are generally imbedded in polyethylene slabs, which moderate the energetic neutrons and increase the overall detection efficiency. Typical sensitivities for a 200 litre drum and a 1000 second counting time are of the order of 1 mg of ^{240}Pu . Note, too, that it is only those nuclides which exhibit spontaneous fission which are detected, along with a small number of neutrons generated by (α , n) reactions. It is thus essential to have some knowledge of the ratios of the various plutonium isotopes before the activity of these isotopes can be assessed. For a low density drum (0.2 g cm^{-3}), this leads to an overall detection efficiency for ^{239}Pu of the order of 60 Bq g^{-1} , well above the acceptable level for material which can be freely recycled.

Active neutron techniques involve irradiating the sample in question with an intense neutron source, either ^{252}Cf spontaneous fission, or an accelerator. This causes a proportion of the uranium and transuranic materials to fission. The external neutron source is shielded and the delayed neutrons generated by the sample are counted using the same detector arrangement as in the passive mode. Typical minimum detectable quantities are of the order of 20 mg of ^{235}U in a 200 litre drum filled with paper and 60 mg of ^{235}U in a drum mainly containing iron. These correspond to activities of approximately 0.04 and $0.0002 \text{ Bq g}^{-1} \text{ }^{235}\text{U}$.

The two modes, active and passive counting, can be combined to give an estimate of the total spontaneous fission and fissionable isotopes in a container.

7.9 Radiochemical techniques

As has been demonstrated in the previous sections there are several nuclides which are difficult to monitor either as a surface contaminant or as a bulk contaminant, for example ^3H and ^{241}Pu . Such radionuclides can often be assessed by radiochemical methods. These are much slower and more expensive than direct methods, because of the much longer preparation time before counting.

The first step is the chemical separation of a sample to isolate or concentrate the desired chemical species. This has two advantages. The more obvious is that it increases the specific activity of the material to be monitored but sometimes the more important one is that radionuclides which emit interfering radiations which could compromise the counting process are removed. The initial step is generally to dissolve the sample. Separation of the chemical species of interest can then be achieved using ion exchange, chelating techniques, precipitation and solvent extraction. Alternatively, for ^3H , the sample can be oxidised and the resulting tritiated water can be condensed, or synthesised into an organic compound for scintillation counting or a gas for proportional counting. The resulting product can then be counted using either the detectors considered earlier or by liquid scintillation counting.

This technique is particularly suitable for low energy β nuclides in that the material to be analysed is dissolved in the scintillator material. This immediately avoids the problems associated with window attenuation. The scintillator material normally comprises a solvent and one or two scintillators. The second scintillator absorbs radiation from the first scintillator and re-emits it at another wavelength. This is often necessary either because the solvent absorbs the light from the primary scintillator and/or because the wavelength of the light from primary scintillator does not match the response characteristics of appropriate photomultiplier tubes. Common solvents include toluene and p-xylene, and common scintillators include 2, 5 - diphenyloxazole (PPO) as a primary and 1, 4 - bis - (2 - (2 - (5 - phenyl - oxazolyl)) - benzene (POPOP) or p - bis - (o - methylstyryl) - benzene (bis-MSB) as secondary scintillators. Primary scintillators are used in concentrations of 4 to 10 g / litre of solvent, with secondary scintillators at about 1 g / litre of solvent.

The light output from the scintillator depends on the energy absorbed from the radioactive decay but is generally reduced by the phenomenon of quenching. The presence of the sample and any carrier reduces the efficiency of the conversion process from ionising radiation to useful light. The degree of quenching is often assessed by bringing an external γ source, such as ^{137}Cs , up next to the sample, measuring the light output, subtracting the light output produced by the sample itself and using this net output to derive a quenching factor. This is then used to correct the sample output.

Liquid scintillation counting is often used for the simultaneous assessment of ^3H and ^{14}C , with the bulk of the ^{14}C signal appearing in a higher window, and the bulk of the ^3H signal in a lower window.

Liquid scintillation counting is capable of very low level measurement because the materials involved, such as the solvents, scintillators, counting vials and photomultiplier tubes, can be obtained with very low levels of radioactivity and also because the system can be effectively shielded. Coincidence techniques are often used whereby a count is registered only when it has been detected simultaneously by two photomultiplier detectors mounted at 90° . This reduces the background due to events in the photomultiplier itself.

The subject is discussed in detail in NCRP report 58 (see Bibliography).

An excellent summary of typical lower limits of detection, and cost per sample (USA 1994) is given in reference 13 for soil samples, from which the following data (Table 24) has been abstracted.

TABLE 24 Radiochemical detection capabilities

for activity in soil

Nuclide	Emission detected	Price to detect 0.4 Bq g ⁻¹ (ECU/sample, 1994)
³ H	β	40 → 110
¹⁴ C	β	95 → 172
⁵⁴ Mn	γ	95 → 172
⁵⁵ Fe	x	60 → 172
⁶⁰ Co	γ	95 → 178
⁶³ Ni	β	60 → 200
⁶⁵ Zn	γ	95 → 172
⁹⁰ Sr	β	75 → 222
⁹⁹ Tc	β	75 → 220
¹⁰⁶ Ru	γ	95 → 172
^{110m} Ag	γ	95 → 172
¹²⁵ Sb	γ	95 → 172
¹³⁴ Cs	γ	95 → 172
¹³⁷ Cs	γ	95 → 178
¹⁴⁷ Pm	β	70 → 172
¹⁵¹ Sm	β	80 → 160
¹⁵² Eu	γ	95 → 178
¹⁵⁴ Eu	γ	95 → 178
²³⁸ U	α	95 → 166
²³⁸ Pu	α	95 → 200
²³⁹ Pu	α	95 → 200
²⁴⁰ Pu	α	95 → 200
²⁴¹ Pu	β	80 → 250
²⁴¹ Am	α	95 → 250
²⁴⁴ Cm	α	95 → 250

These prices demonstrate that radiochemical analyses are inevitably much more expensive than direct methods. Their main use is likely to be in the finger printing process.

8 Some examples of decommissioning programmes

These examples are derived from UK experience of decommissioning monitoring. Some are reported as well defined specific cases whilst others are the aggregated experience from a number of similar situations.

8.1 Removal of a pneumatic tube transport mechanism

This mechanism had been used to transport specimens rapidly from a materials test reactor, where they had been neutron activated, to analysis facilities elsewhere on site. Use of the mechanism had ceased several years before. This was a relatively attractive project because

- : the route of the pipe was well known.
- : the pipe was relatively close to ground surface.
- : the pipe ran mostly through areas where excavation was easy, ie, under open grass areas.
- : the material, copper, is relatively valuable.
- : the pipe had not been used for several years and an assessment of the possible contaminants showed that ^{60}Co was the only likely problem.

Monitoring aspects

It was possible to perform all the measurements using only one instrument, a 51 mm x 51 mm sodium iodide detector connected to a scaler ratemeter. The initial measurements involved walking the route of the pipe with the instrument held at ground level in the ratemeter mode. This was intended to identify the presence of any hot spots, ie, the presence of any really active particles within the pipe. Limits of detection were estimated to be approximately 40 MBq, when buried at a typical depth of 1 m. None was found.

The ground was then excavated to reveal the pipe. The monitor was then scanned along and in contact with the pipe, again in the ratemeter mode. This was intended to identify ^{60}Co particles in excess of 5 kBq. The pipe was then disconnected, the open end sealed on both the off cut and the remaining section and the 2 m length lifted out and placed on tressels. Agreement had been reached with the relevant regulatory body to allow sentencing in 2 m lengths. This was performed by setting the instrument to integrate for 100 seconds and then moving the detector back and fore over the full length of the pipe. The result was then compared with the local background. The acceptable maximum level was set at 0.4 Bq cm^{-2} which corresponded to a net count of 1200 above a typical background of 7000. Problems which were encountered were mainly caused by having to work in areas of enhanced gamma dose rate, generated by other site facilities. This caused the background count to go beyond 9999, the maximum that the instrument could display. In these circumstances the lowest accessible background area was selected and the time taken for the instrument to reach 9999 noted. This was displayed automatically. The measurement was then repeated with the pipe to be monitored and the time to 9999 again noted. The net count rate was then calculated and compared to the release value. Obviously the statistical uncertainty was increased using this technique but no operational problems were encountered. The criticism can be levelled that it might have been better to use more sophisticated equipment but the decision was made to use the simple unit because:-

- (1) it is tough, waterproof, and had no external calibration controls.
- (2) spare units were available.
- (3) the workforce found it easy to use and had confidence in it.
- (4) it had a history of use in such operations.
- (5) the radiation characteristics were well established.

8.2 Clearance of asbestos insulation

Decommissioning of a power station generally involves the removal of large amounts of asbestos, which has to go to some form of approved waste disposal site. That site may, or may not be, licensed to handle very low activity radioactive waste. In any case assessment of activity of each bag of waste is normally performed.

This is a relatively simple task because:-

- (1) the likely contaminants can be established and the finger print is likely to contain a significant proportion of energetic γ emitters. In one case the γ activity was assessed as 14% of the total.
- (2) the waste will be in sealed bags, which can be weighed.
- (3) the density of the material will be low which minimises potential problems from self absorption.
- (4) the outside of the bags will be clean.
- (5) the bags can be fitted in a standard small items monitor, giving virtual 4π monitoring using thick plastic scintillator, under efficient shielding.

This equipment, used with a counting time of 100 seconds, gave reliable detection of bags with a mean activity of energetic γ emitters in excess of 0.07 Bq g^{-1} and hence of a total activity in excess of 0.4 Bq g^{-1} .

8.3 Monitoring of the inside of fume cupboards and glove boxes

Fume cupboards and glove boxes are normally constructed from impervious materials such as stainless steel and plastic laminates, which are easy to clean and monitor. Experience shows that dismantling and decommissioning generally involves the following procedure.

- (1) Find any historical data on use, nuclide content and monitoring results. It is important not to trust very old data as records may not be complete.
- (2) Perform an initial probe survey using a probe which is suitable for the expected nuclide mix. In many cases a dual α and β proportional counter is the most suitable, with a dual α and β scintillation probe acceptable if low energy β emitters are unlikely.
- (3) Any areas which are obviously above background should be monitored using a wipe. This will provide more robust information on the removable activity.
- (4) Clean any areas where the apparent activity is not much above the acceptable level for release. Confirm by probe and swab.
- (5) Identify areas where monitoring is impossible in the assembled structure such as joints and overlaps. These areas may conceal relatively high levels of activity which can be released during the dismantling process. One option is to cut these areas off before dismantling, if possible, or monitor them as soon as access is possible.
- (6) Sentencing the resulting material based on the allowed averaging area.

All measurements should be recorded which will give confidence to the relevant regulatory bodies and to the purchasers. This process will also help identify the point of failure if, further through the process, an article is found to have a different level of surface activity from that expected.

The process can also be used in situations where there is a mix of nuclides, including nuclides which are very difficult to monitor directly, such as ^{63}Ni and ^3H . Radiochemical analysis, γ spectrometry and activation analysis can be performed to a range of samples. If a relatively stable finger print is obtained then a release level limit can be calculated for the probes and wipe counters employed.

One important problem which can be anticipated is that a probe may indicate that no α activity is present while a wipe may indicate significant levels. Monitoring by probe generally involves a larger surface to detector separation. Probe windows are thicker and are protected by a denser grill. The result of these differences is that alpha particles emitted from a surface which has significant self absorption because of the presence of grease, for example, are likely to be recorded in the beta channel rather than the alpha.

8.4 Monitoring of equipment used in the handling of ^{226}Ra

Historically large quantities of ^{226}Ra were used in luminising and the preparation of sealed sources. The crux of the problem, and the difference from the majority of other decommissioning exercises, is that ^{226}Ra decays to ^{222}Rn , which is a noble gas with a half life of 3.8 days. The decay products of ^{222}Rn , involving such long half life nuclides as ^{210}Pb , can thus be found a long way from the nominal work site, leading to long lengths of, for example, contaminated air trunking. It is often impossible to find out exactly what operations took place in such facilities and what quantities of ^{226}Ra were used. Large quantities may have been spilled in a fume cupboard and drawn into the ventilation system. Parts of the facility may have been reconditioned. Hence the absence of activity in one part cannot be taken to indicate the lack of activity in another.

8.5 Monitoring of scaffold boards and tubes

Decommissioning operations also involve the clearance or sentencing of equipment used in the operation itself, rather than the products of such operations. Typical of these are scaffold boards and tubes. For work involving radioactivity, boards should not be made of wood, which is very difficult to monitor, but of some impervious material, such as aluminium. Scaffold tubes should also be capped before entry to the site so that the interior can be guaranteed uncontaminated.

Monitoring of such large areas is best done using a very large array of gas flow proportional counters. The counters should be wider than the width of the greatest object to allow monitoring of the sides of boards. The length of the array should be chosen by balancing the cost of the equipment against throughput. For a large operation, where regular monitoring of the equipment is desirable, a full length double (top and bottom) array may be justified. For smaller operations monitoring in quarters, ie, top and bottom face for one half, and then repeated for the other half, may be the most economic.

The equipment can be set up to monitor both alpha and beta activity but inevitably for any object which is not flat the alpha measurement will be inefficient. The length of each monitoring element can be chosen to give monitoring areas corresponding to those permitted in relevant regulations. Each element can be connected to an alarm, which will clearly indicate to the operator which, if any, segment is contaminated.

Scaffolding boards and tubes are relatively light and 2 people can manipulate them easily. Consideration should be given as to whether automatic identification by bar code has merit although the rough usage involved in demolition and clearance may well preclude it.

8.6 Monitoring of steel beams

Large quantities of steel beams can be expected during the demolition of any large structure. In some areas it will be easy to calculate that activation is insignificant and that surface contamination is unlikely. In other areas, particularly with early reactors where significant neutron exposure of steel outside the pressure vessel and biological shield may have occurred, then activation is possible.

Steel beams are very heavy and in most circumstances it will be better to fix the beam and move the monitoring equipment along it. Suitable techniques would involve sodium iodide γ spectrometry to identify activation products such as ^{60}Co , combined with gas flow proportional counters mounted in an array giving a good view of the whole surface of the beam for the monitoring of contamination.

Rates of monitoring could be up to one metre per minute using large area proportional counters and large volume sodium iodide detectors.

For this sort of operation the beam has to be uniform, which means that rivetted and welded joints will have to be cut off and treated separately. Again the problem arises that overlaps are difficult to monitor if the possibility exists of unacceptable levels of activity within the overlap.

Another problem which may be encountered is if the steel has been painted after the facility has been in use. Paint will mask low energy beta emitters very successfully (see section 6.4) and, in common with many circumstances, removal of paint may be necessary in order to allow clearance monitoring. However, in areas where fingerprinting indicates that energetic beta emitters will generally be present it may be possible to set an initial reject level on the basis of the calculated response for the monitor at the limiting level with the maximum thickness of paint expected. Beams which fail this criterion can be stripped and re-monitored. Build up of data from this process may allow the relaxation of the initial reject level for areas where there is no reason to suspect gross non uniformity of deposition.

8.7 Decommissioning of a gaseous diffusion plant

Gaseous diffusion plants are very attractive targets for decommissioning operations. They tend to use large quantities of potentially valuable materials such as aluminium, nickel and stainless steel. Activation is not a problem. The radionuclides present are generally well known and monitoring during repair and maintenance operations will give a useful historical background. Plants tend to be composed of large numbers of very similar units. Hence it is possible to start the process with a reasonable expectation of the problems and to build up knowledge relatively quickly.

Monitoring of large simple surfaces is easy, either by hand or automatically. The main problem in components like process vessels is the presence of welds. The material being processed is gaseous and tends to penetrate welds. Cleaning on shut down is normally very effective for exposed surfaces but not for welds. Hence it is possible to have a clean surface covering a contaminated volume. The implication of this is that welds must either be split and monitored or cut off and treated separately. Dealing with valves and other small components is more difficult. An attractive technique is to cut the components up to lay surfaces open to wet decontamination in a purpose designed plant and then melt the resulting cleaned objects. The furnace off gases can be filtered and treated to minimise discharges to the environment and the resulting ingots monitored for sentencing. This process has many advantages.

Monitoring of the ingots can generally be done by performing a limited programme of fingerprinting by radiochemical methods. Data from this process can be used to calculate the response of a low background alpha and beta counter based on samples of fixed geometry. Counting times of

the order of one thousand seconds will generally confirm whether the residual activity in the sample allows free release, ie, it is less than 1 Bq g⁻¹ total for α emitters.

The main cost of this process is the installation of the melting plant. The monitoring costs are relatively trivial. Even if the material produced is not fit for free release the volume has been reduced and burial will be followed by a much slower release to the environment.

8.8 Monitoring of earth contaminated by leakage from a fuel storage pond

This was a particularly simple situation. Samples of the soil were subjected to gamma spectrometry and radiochemical analysis. The dominant nuclide was found to be ¹³⁷Cs and agreement was established with the regulator that the waste was sentenced on that basis. Measurement was performed using a simple 50 mm x 50 mm sodium iodide scintillation detector connected to a ratemeter. Calculation (see 7.3.1) indicated that a response of approximately 200 s⁻¹ Bq⁻¹ g was to be expected with the probe held close to a large volume of the material. Final calibration of the process was achieved by filling the loader bucket with a well mixed sample with an average activity established by gamma spectrometry. Measurements were performed with the scintillation probe on top of the bucket, where there was no shielding, and in contact with the sides of the bucket, where there was shielding by the steel wall.

Maximum acceptable count rates for unrestricted release were calculated using these data for measurements on top and in contact with the side of the bucket.

8.9 Monitoring of the internal and external activity of scaffold poles and process piping

Normally scaffold poles have capped ends welded into place. However, on some older sites, open poles can be found. These are difficult to monitor either for reuse off site or for recycling because of the difficulty of confirming that there is no contamination inside the pole. It is possible to some extent to pull a wipe through the pipe using a wire and then to monitor the wipe. However this is rarely deemed acceptable by licensing authorities because of the difficulty of interpretation of the apparent activity on the wipe. Pipes can, of course, be cut into shorter sections which are easier to monitor or, indeed, cut open length ways. Both these processes are laborious and mean that the pole becomes scrap metal.

It is possible, however, to monitor a 6 m length of pole as one unit for alpha activity using the ion chamber detector described in section 6.1.2.5. In a current design the pole is loaded into the machine, air sucked through and over the pole and the current generated measured. The end of the pole can then be capped, or a gag valve closed around the pipe, and the process repeated. From this process two currents can be determined, one generated by any activity within the pipe and the other by activity on the outside of the pipe. An appropriate calibration factor can be used to derive equivalent total activities in Bq, with a lower limit of detection of the order of 20 Bq. The process only really works with poles which are reasonably clean, in the sense of not having obvious loose surface deposits, but does offer rapid monitoring, with a time of approximately 2 minutes per piece, with a low limit of detection and acceptable precision.

The same technique can also be applied to process pipework, valves and, with appropriate adaptors, ducts.

8.10 Monitoring of cable

Decommissioning often leads to the generation of large volumes of scrap electric cable, which have significant value. Frequently the decommissioning contractor or site operator is confident that the cable is neither neutron activated nor significantly contaminated on its surface but it is necessary to demonstrate this to the regulatory authorities.

Electrical cable is not particularly easy to handle as frequently the removal of a cable tray will yield cables of varying diameter and stiffness, which cannot be packed easily into a compact mass. Monitoring, then, frequently involves dealing with relatively small boxes or plastic bags containing of the order of 25 kg of cable and having a density not much greater than 1 g cm^{-3} . This low density can be compared to a typical density of low power cable of approximately 4 g cm^{-3} , which implies that the cable occupies only about 25% of the available volume.

Monitoring of such material can be performed, normally using a large sodium iodide scintillation detector. The monitor should be scanned rapidly round the container, to check for hot spots, and if none are found, held in contact with the centre of the top surface of the scrap material. The count rate should be noted and an appropriate background value, based on a sample of clean cable, subtracted to yield a net count rate due to contamination.

The response of the monitor to a typical sample can be calculated by determining the mean solid angle subtended by the detector. To achieve this, the sample can be considered as a stack of discs of identical thickness with the detector in contact with the centre of the top disc. To simplify matters the detector can be considered as being a sphere of the same volume as the genuine detector. As an example, for a 51 mm diameter, 51 mm deep crystal the radius of the equivalent sphere is 29 mm, and the area of a sphere of this radius is 2640 mm.

Based on this for a detector area A with its centre at a distance d from a disc radius R then the fractional solid angle d_i subtended by the detector =

$$\frac{A}{4\pi R^2} \left[\log \left(1 + \frac{R^2}{d^2} \right) \right]$$

Attenuation for that disc can be calculated using the mean path length between the detector and the disc.

Using the same symbols as before gives mean path length x =

$$\frac{2}{3R^2} [(R^2 + d^2)^{3/2} - d^3]$$

This value should then be multiplied by the average density and inserted into the standard equation as ρx , ie,

$$I = I_0 \exp \left(-\frac{\mu}{\rho} \cdot \rho x \right)$$

The value of μ/ρ can be estimated from a consideration of the gamma component of the fingerprint and the atomic number of the absorber, typically copper, using one of the standard tables.

If each disc is assumed to emit one photon per second per unit area then the number of photons striking the detector from disc i is:

$$4\pi R^2 d_i \cdot \exp \left(-\frac{\mu}{\rho} \cdot \rho x \right)$$

These values can be summed for the stack of discs approximating to the complete volume of the bag or box. The mean detection efficiency of the detector can be estimated from the response of the detector in $\text{s}^{-1} \mu\text{Gy}^{-1} \text{h}$ derived from its calibration at an appropriate energy.

From this, the response of the detector can be determined, at least approximately, in terms of counts per photon emitted per unit volume which can then be expressed as a count rate corresponding to the limiting activity in the bag.

There are obvious uncertainties and approximations in this process but it can often be sufficient to allow recycling of the large volumes of cable which have been in an active area but which can reasonably be expected to be uncontaminated.

8.11 Monitoring of uranium contaminated plastic covered steel floor plate

Steel floor plate is frequently found in buildings on nuclear licensed sites. On older sites these floors are occasionally covered with plastic flooring material which has sometimes been laid over, what was at the time, an acceptable level of contamination. One example concerned a building which had been used for work with depleted uranium.

The aim was to remove the waste without removing the plastic floor covering, which was approximately 2 mm thick and of unit density and which was extremely well stuck down. It is obviously impossible to monitor the uranium alpha emissions through the significant covering. However ^{238}U will normally be found in equilibrium with $^{234\text{m}}\text{Pa}$, which is an energetic beta emitter with a maximum energy of 2.28 MeV. There were two aspects to the task.

The first aspect was to connect a large area anthracene based scintillation detector to a gated ratemeter, ie, one with a selectable counting window. A contaminated sample of the plate was obtained and the detector set up so that its minimum detectable activity was as low as possible, which in this case involved narrowing the window to reduce the background upto the point where a significant reduction in the signal was observed.

The second aspect was that this detector was then calibrated using a uranium source of known activity covered by a layer of material of the same mass per unit area as the floor covering.

In this manner a monitor was produced which had a background of approximately 2 counts per second and a response to 1 Bq cm^{-2} of ^{238}U in equilibrium of 3 counts per second. This allowed disposal as the surface of the plastic material was uncontaminated and the mass per unit area of the plate was 3 g cm^{-2} , giving compliance with the suggested 1 Bq g^{-1} limit.

9 Equipment costs

Equipment costs are an important part of the equation when comparing the cost effectiveness of attempting free release of material to the costs of on site storage or disposal.

Some examples are given below. A good general rule is that equipment which is in use for general radiation protection applications is relatively inexpensive whereas equipment which is designed for a particular application is expensive, especially if the manufacturer considers that further sales are unlikely. Additional cost considerations are given in Annex A.

Instrument	Typical cost, 1997 ECU
1 Simple large area β contamination probe and ratemeter	1,800
2 50 mm x 50 mm sodium iodide scintillation probe and ratemeter with simple pulse height selection	4,000
3 Large area, refillable, hand held α/β proportional counter	3,000
4 Simple γ spectrometer using a robust 50 mm x 50 mm sodium iodide scintillation detector and multi channel analyser in a single, hand held, unit.	15,000
5 High efficiency intrinsic germanium detector and multi channel analyser with emulation software, as a drum monitor	100,000
6 Set of proportional counters for scaffold board monitor and control electronics	75,000
7 α monitoring unit for tubes, etc.	Under development No price available
8 Bag monitor for the measurement of gross γ activity	50,000
9 Small article monitor for small pieces of pipe and individual components	20,000

10 Bibliography

The following references offer general information on the measurement of radioactivity. They will be found extremely useful in gaining understanding of the physics of measurement of radioactivity and in the design of a monitoring programme.

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- 12 Conversion coefficients for use in radiological protection against external radiation. ICRP Publication 74, Pergamon, (1996).
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12 Figures

- 1 Photon energy response of NE Technology IDP6AD.
- 2 Photon energy response of the Berthold LB6357 xenon filled detector.
- 3 Photon energy response of the Berthold LB6358G Butane detector.
- 4 Photon energy response of the Mini Instruments 1000 EP15.

Annex A

Cost Considerations

A1 Cost considerations

The costs of monitoring during decommissioning for disposal or recycling of materials will need to be assessed on a case by case basis, but there are some general features which will be amenable to a systematic analysis of costs, so they can be minimised.

A2 Radionuclide inventory

The initial assessment of the radionuclide inventory in materials on a site will be a cost which will vary from site to site, and will require a mixture of monitoring and a study of archive documents to establish the extent of the inventory. The monitoring will need to be specialised, and will require a mixture of hand held instrumentation and laboratory analysis of samples to establish radionuclide fingerprints. In order to minimise the number of man-hours required to do monitoring with hand-held equipment, it will be an advantage to have the most sensitive instrumentation available. However, the prospects for making large cost savings by using different monitoring systems or techniques are likely to be small compared to other costs. So, despite the inventory assessment varying from case to case, the costs for a particular site will be relatively fixed with respect to changes in monitoring equipment. The main opportunity for cost savings will come at the next stage when significant quantities of materials are being processed for recycling or disposal.

After the initial assessment of a radionuclide inventory has been carried out, decisions can be made for disposal or recycling materials. In respect of considerations of the cost of monitoring, there are two significant decisions that are amenable to a systematic analysis of the costs involved: first, whether to use an automated or manual system for monitoring materials, and second, decisions on the type of monitoring equipment to be used. Both these decisions can be analysed using cost-effectiveness techniques.

A3 Cost-effectiveness analysis

The technique of cost effectiveness analysis involves the examination of options to define a ratio of benefits, b , to costs, c . The most cost effective option is that which maximises benefits for a fixed cost, or which minimises costs for a fixed benefit. In respect of decommissioning and recycling, the benefit is likely to be a quantity of material sentenced either to disposal or to recycling (kg/ECU). Hence a cost effectiveness analysis should demonstrate the options which can sentence the most material for a fixed cost or which can minimise the cost of sentencing a fixed amount of material.

The technique of cost effectiveness analysis is fundamentally different from cost benefit analysis, with which it is often confused^{1,2,3}. Cost benefit analysis seeks to optimise (ie, maximise) the net benefit minus costs ($b-c$) of particular options. This requires a much broader treatment of benefits in monetary terms, which can be both difficult and contentious⁴.

A4 Automated versus manual monitoring

The decision to use automated or manual monitoring will require a detailed analysis of the amount and properties of the material to be processed, and the availability of suitably trained staff for monitoring. If there are large amounts of relatively uniform material that will have to be processed in

some way before removal from the site, the intuitive arguments for an automated monitoring process are strong. Conversely, if there are smaller amounts of very disparate materials with a variety of shapes and sizes, manual monitoring by skilled staff may be more appropriate. However, if such material will have to be processed in some way before its removal from the site, (eg, because of regulatory requirements) then the arguments for an automated monitoring system will become strong, especially if there are large amounts of material. This can be shown in terms of a schematic example.

Let us assume a site for decommissioning has been designated and the assessment monitoring and archive studies have been carried out to establish the likely radionuclide inventory. (This is then a fixed cost, whatever subsequent monitoring regime is adopted.) This inventory suggests that some material may be recycled but some other material may need disposal either using general disposal techniques or special disposal at sites suitable for radioactive wastes.

Let the total mass of material be M kg and the process diagram given in Figure 1A illustrates how the cost effectiveness of monitoring regimes can be calculated.

The schematic diagram demonstrates that automatic or manual monitoring may be the only significantly variable cost in the overall process. Assuming conservation of mass during the three different process chains:-

- I Processing; automatic monitoring; disposal
- II Processing; manual monitoring; disposal
- III Manual monitoring; processing; disposal

and that the amounts of material eventually sentenced to recycling, disposal or radioactive waste are not significantly different at the end of the three process chains then a straightforward summation of costs is revealing. For example:

Process I will have a total cost $(X + Y - R^* + D + S)$ ECU,

Process II will be $(X + Z' - R' + D' + S')$ ECU and

Process III will be $(Z'' + X'' - R'' + D'' + S'')$ ECU where $X'' = x_1'' + x_2'' + x_3''$

where X (or x) denotes the cost of processing material before its removal from the site.

Y denotes the cost of automated monitoring material (ie, after processing).

Z denotes the cost of manually monitoring material (before or after processing).

R denotes the cost of recycling material (this should actually be a net income because of the value of recycled materials).

D denotes the cost of disposal of non-radioactive materials.

S denotes the cost of disposal of radioactive waste.

The assumption that amounts of material eventually sentenced will not vary significantly between I, II and III means that

$$R = R' = R'' ; X = (x_1'' + x_2'' + x_3'') = X''$$

*Recycling should be a net income

$D = D' = D''$; $Z' = Z''$; and

$S = S' = S''$

and the costs of process II and process III are therefore equal. The only difference between process I and process II/III is then the cost of automated and manual monitoring. Hence, what decides the cost effectiveness of the overall scheme is the cost of the monitoring process.

Taking a more “real world” approach, the above assumptions are naive and it is unreasonable to assume that the masses of materials eventually sentenced to recycling, disposal or radioactive waste are independent of the monitoring process. For example, a manual monitoring procedure may not reveal the extent of radioactive materials in concealed or obscure places, which subsequent automated monitoring could discover. Conversely, the processing of material may obscure some radioactive objects in an automatic monitoring regime. Such difficulties are best considered by experts/regulators involved in the process, but this analysis shows that the cost of the monitoring process could actually be a major contributor to the overall cost-effectiveness of an operation, and can equally have an effect on the amounts of material eventually sentenced to recycling, disposal or radioactive waste. This may be particularly sensitive for recycling which is a negative cost (ie, an income) and for disposal which is likely to be a major cost, especially for radioactive materials.

A5 Selection of monitoring equipment

Given that cost effectiveness analysis has been used to decide whether to use automatic or manual monitoring should be used these two techniques can also be subject to a separate cost effectiveness analysis. It is straightforward to assess that more sensitive equipment can minimise the time required for manual monitoring and also improve the efficiency of automated monitoring because a larger amount of material can be processed for a fixed time. Such analyses could be complex and, in practice, will depend on the nature of the decommissioning project. For example, the equipment required to deal with materials with potential surface contamination will be very different from that required to monitor materials suspected of containing activation products. However, a cost effectiveness analysis designed to assess the maximum amount of material for a fixed cost, or the minimum cost to monitor a fixed amount of material, will be revealing in either case. For any sizeable project involving large amounts of material, a cost effectiveness analysis will reveal that a more sensitive instrument will be able to monitor larger volumes of material more quickly, or minimise the amount of time for manual monitoring.

It should be noted also that there is no straightforward relationship between the sensitivity of the instrument and its cost; some very sensitive instruments can be relatively inexpensive.

A6 Conclusions

The schematic analysis of costs given above shows how significant the cost of monitoring can be in a decommissioning project if the regulatory regime demands that all material must be subject to monitoring before leaving the site. This is expected to be the case where there are significant quantities of contaminated material on site and the arguments for setting up an automated monitoring system on site are strong. If however the assessment of radionuclide inventory reveals only small amounts of contaminated materials, such a rigorous regime may not be necessary and there are arguments for a less expensive manual monitoring scheme.

A7 References

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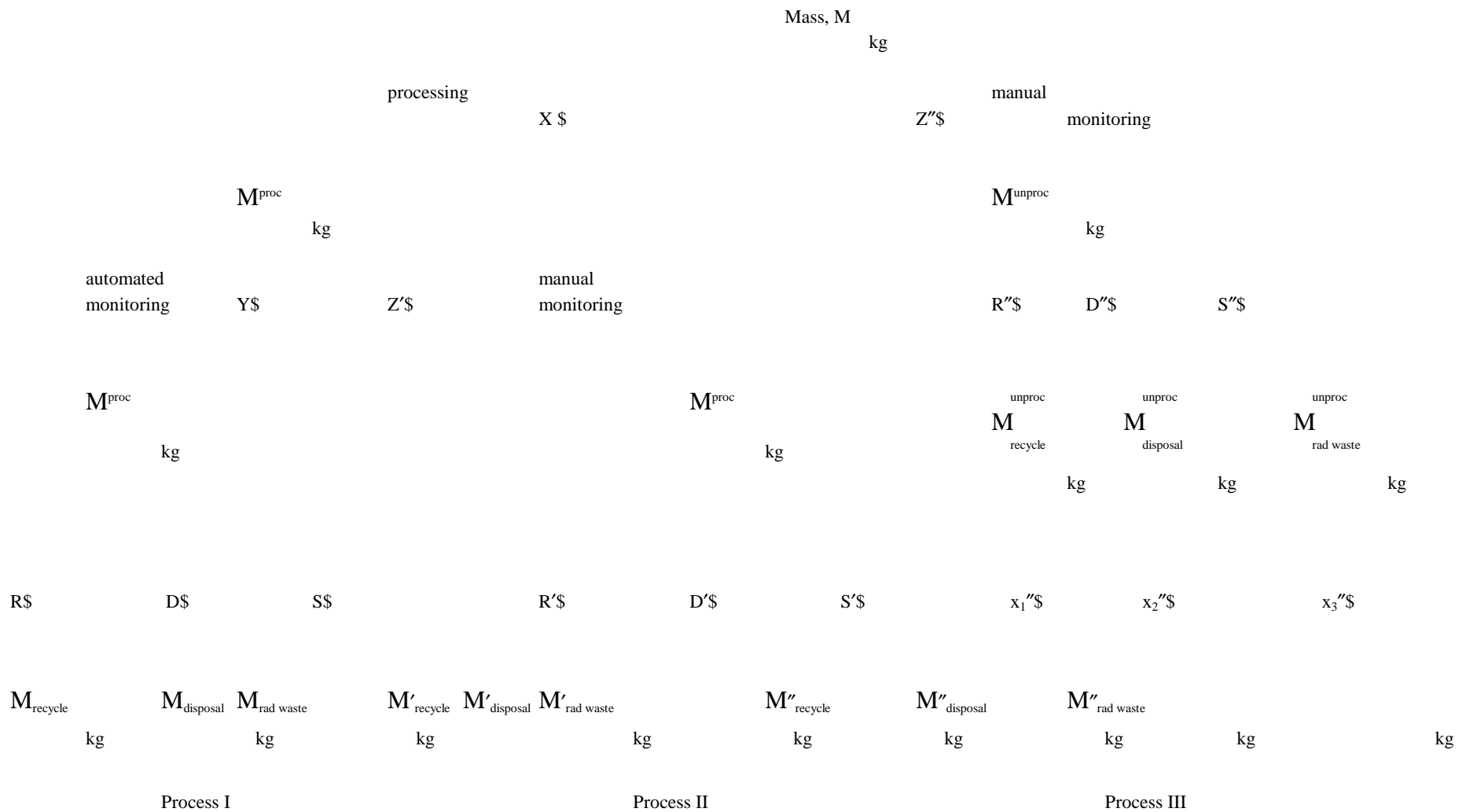


Figure 1A Schematic diagram of decommissioning processes and costs