

**QUESTION 4**

You are the radiation safety officer for a radiopharmaceutical laboratory. A laboratory worker reports to you that he accidentally boiled to dryness a beaker containing 15 mCi of  $^{131}\text{I}$  in a room that is  $5\text{ m} \times 5\text{ m} \times 5\text{ m}$ . He was not using the available fume hood. Assume that the  $^{131}\text{I}$  is instantaneously vaporized and uniformly distributed at  $t = 0$ , and the worker is in the room for one hour after  $t = 0$ .

**GIVEN**

- Breathing rate,  $B = 1.2\text{ m}^3\text{ h}^{-1}$ .
- Room ventilation rate,  $F = 100\text{ m}^3\text{ h}^{-1}$ .
- Fractional thyroid uptake of  $^{131}\text{I}$  from blood = 0.3 (remainder goes directly to excretion).
- Respiratory tract deposition fraction = 0.75. (Assume 100% goes instantaneously to blood).
- Committed dose to the thyroid =  $5.5\text{ rad } \mu\text{Ci}^{-1} \text{ }^{131}\text{I}$  deposited in the thyroid.

**POINTS**

- 30**    A.    Calculate the worker's thyroid uptake of  $^{131}\text{I}$  and the committed dose due to that uptake. **Show all work.**
- 20**    B.    In addition to performing bioassay on the worker, list five actions you will take following the accident. **Number your responses. Only the first five responses will be graded.**